

# Techniques for Synthesizing Polymer Particles

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May 31, 2007



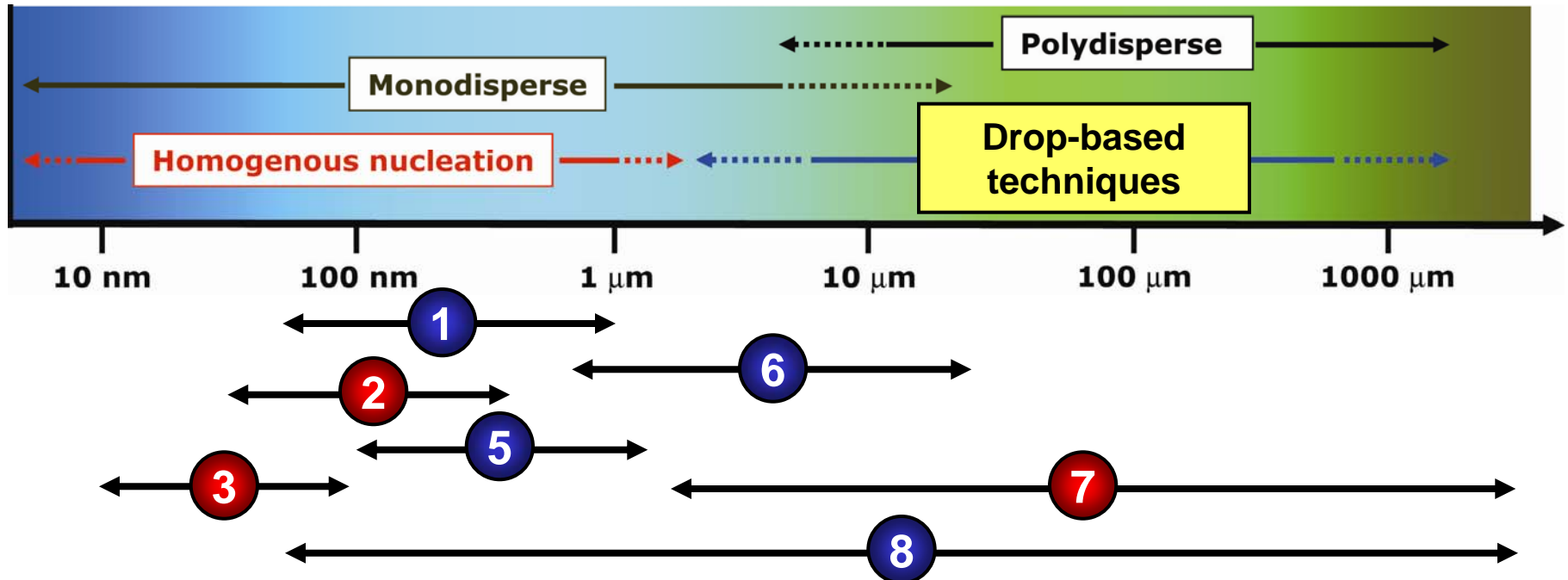
Harvard  
School of Engineering  
and Applied Sciences

# Current techniques

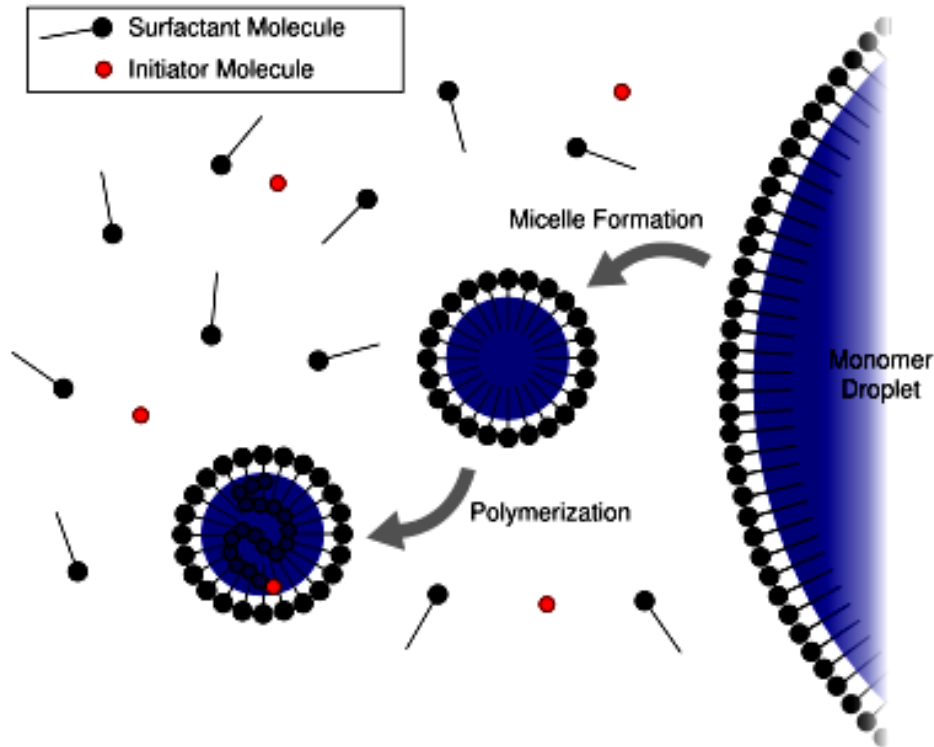
1. Emulsion polymerization
2. Miniemulsion polymerization
3. Microemulsion polymerization
4. Precipitation polymerization
5. Dispersion polymerization
6. Suspension polymerizations
7. Seeded polymerization

## Basic considerations

- A batch process
- Radical polymerization
- Monodispersity: <30%



# (Macro)emulsion polymerization

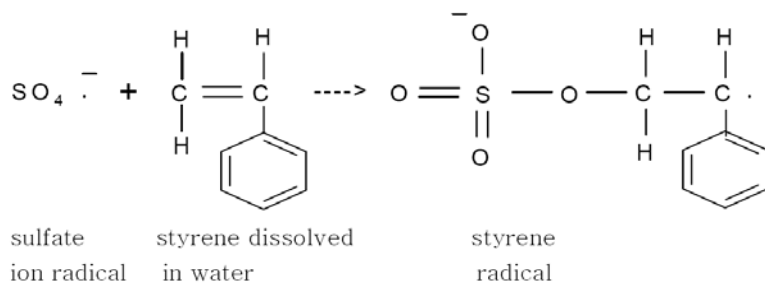


## Features

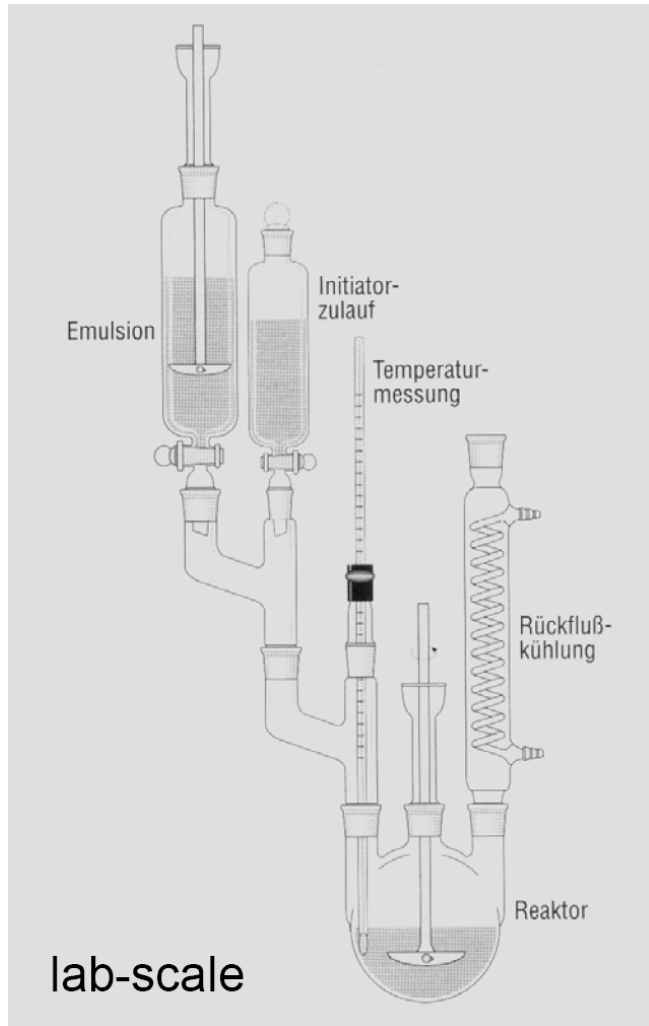
- Heterogeneous system
- Nucleation site: water (in micellar or homogeneous)
- ~20nm – ~1μm
- Monodisperse
- Polymerization time: fast
- High MW
- Surfactant-free is possible

## Components

- Monomers
- Water
- Water-soluble initiator
- Surfactant

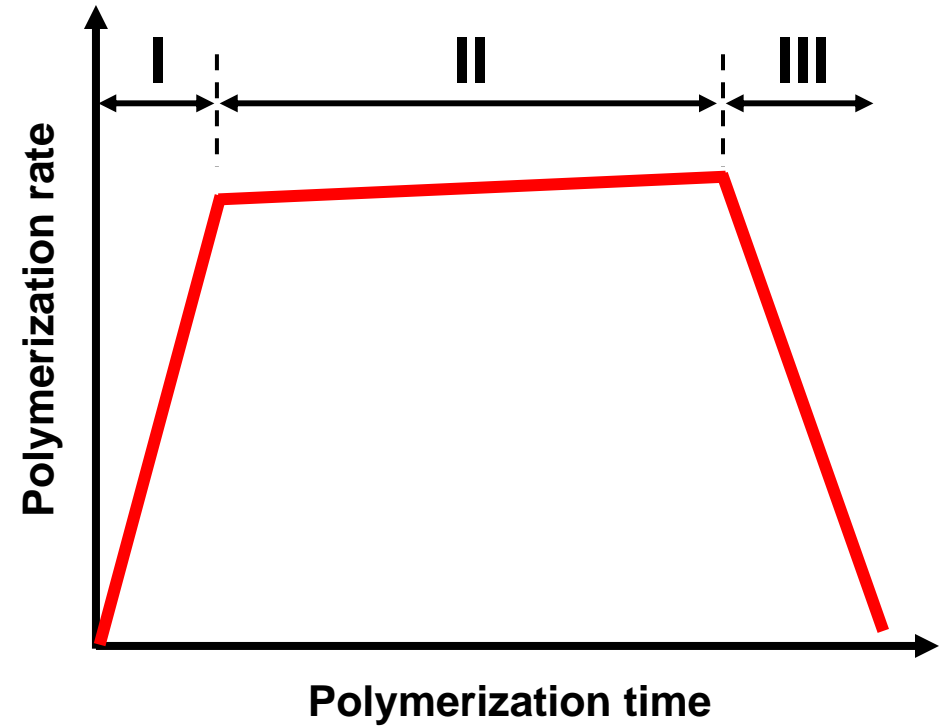
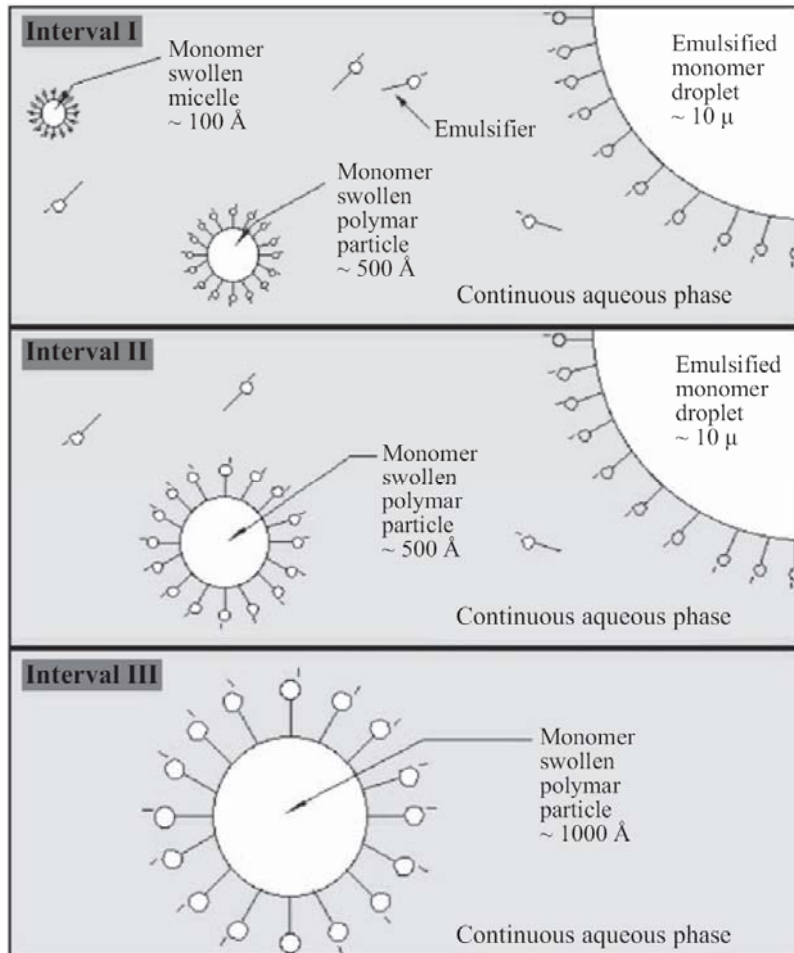


# Reactors





# Stage of emulsion polymerization



## **Interval I**

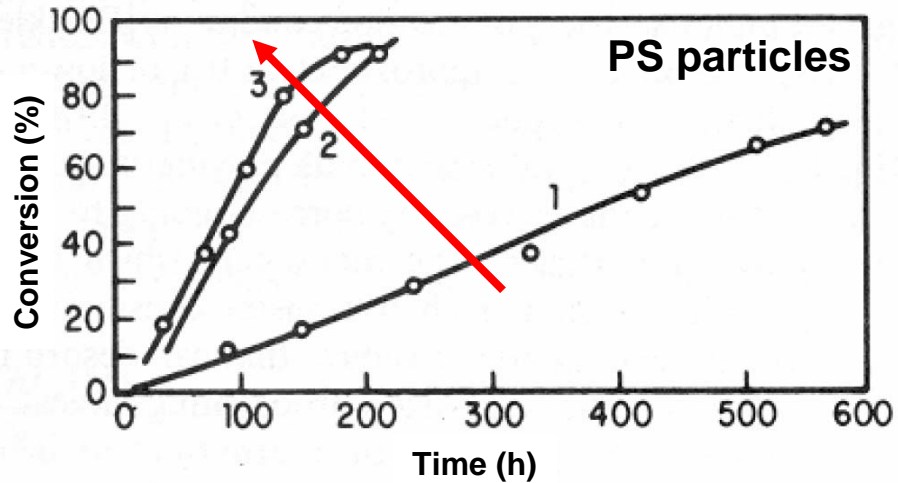
- **2-15% conversion**
- **Radical capturing and # of particles increase (rate increase)**
- **As particle grow, more surfactants adsorb**
- **Surfactant concentration falls below CMC**
- **All the surfactant adsorb to the particles at the end of interval 1**

## **Interval II**

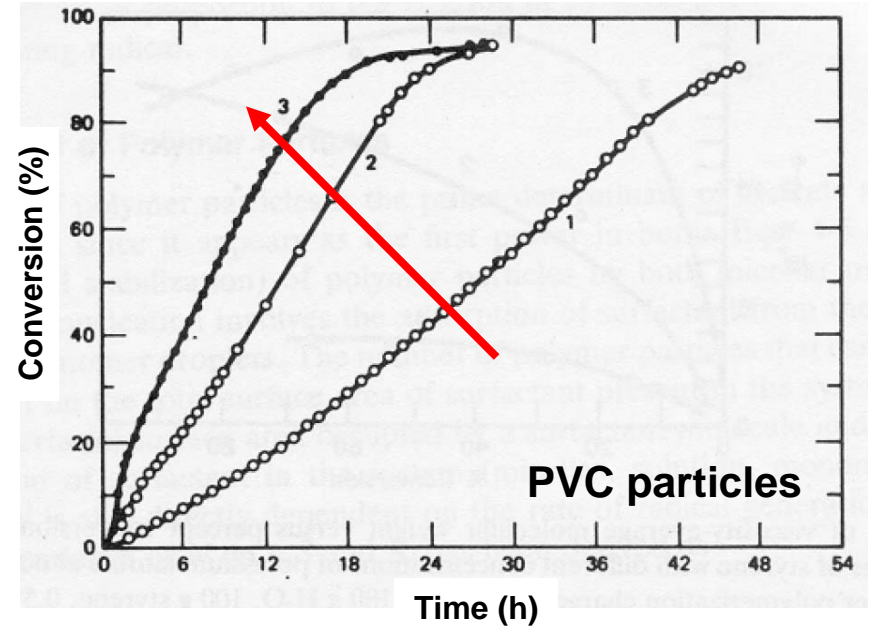
- **# of particles remains constant**
- **Slight increase due to Trommsdorf effect**
- **Interval II ends when all the monomer droplets disappear**

## **Interval III**

- **# of particles remains constant**
- **[M] decreases (rate decrease)**



- Increasing surfactant
- Increasing # of particles



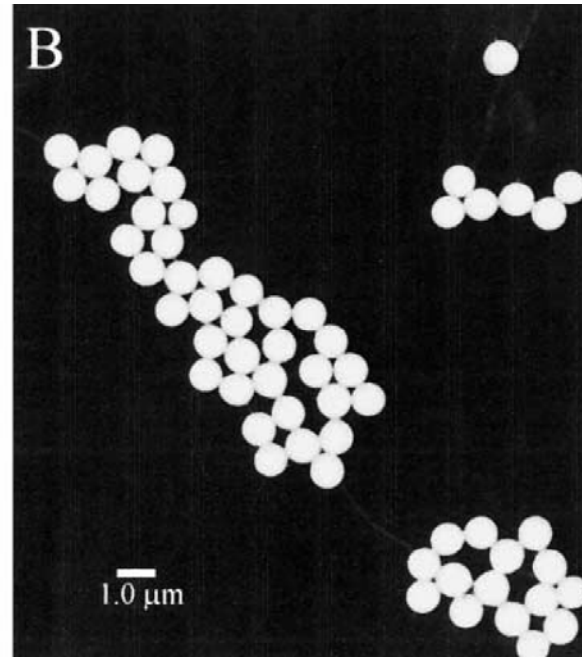
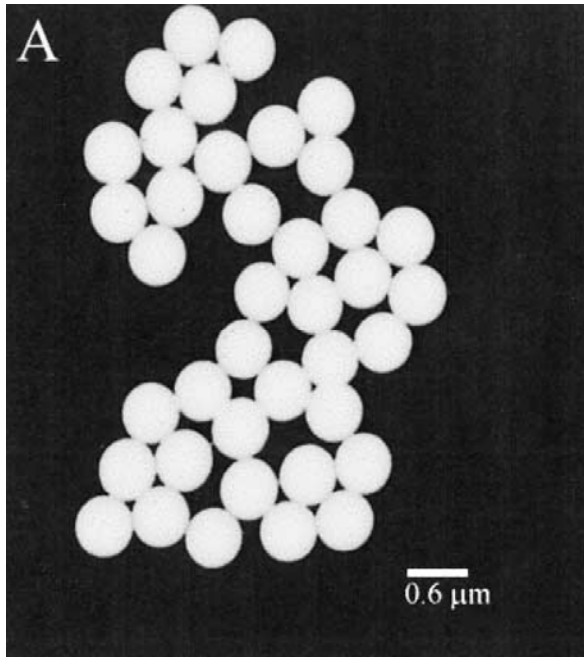
- Increasing initiator concentration
- Increasing # of particles

$$R_p = k_p [M]_p N_p \bar{n} / N_A$$

$$DP = r_p / r_i = N k_p [M] / R_i$$

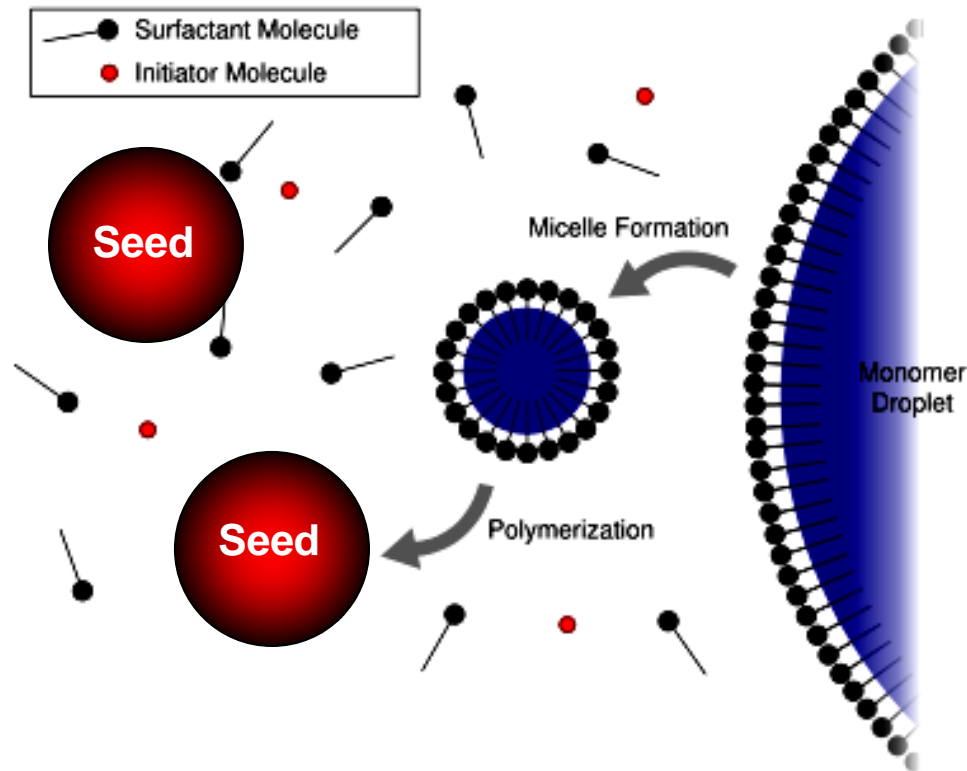
$$N_p \propto [\text{initiator}]^{0.4} [\text{emulsifier}]^{0.6}$$

# Polymer particles



PS particles

# Seeded emulsion polymerization



## Features

- Heterogeneous system
- Nucleation site: water
- Bigger than seed particles
- Monodisperse
- Good for controlling morphologies

## Components

- Seed polymer particles
- Monomers
- Water
- Water-soluble initiator
- Surfactant

# Thermodynamic effect

Thermodynamic control



Core-Shell



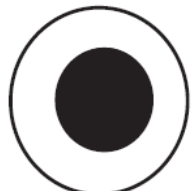
Moon-like  
3rd quarter



Moon-like  
2nd quarter



Moon-like  
1st quarter



Inverted Core-Shell

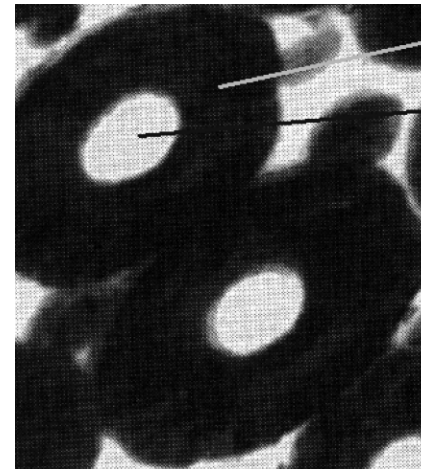


Seed  
polymer

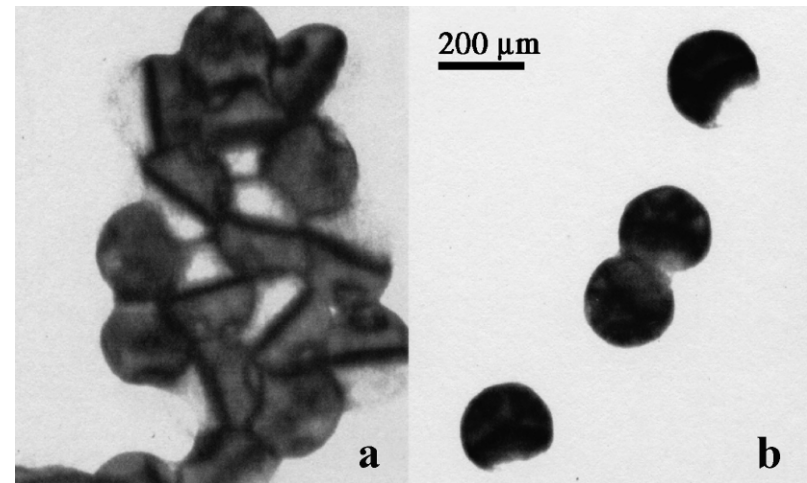


Second stage  
Polymer

$$\Delta G = \sum_i \gamma_i A_i$$

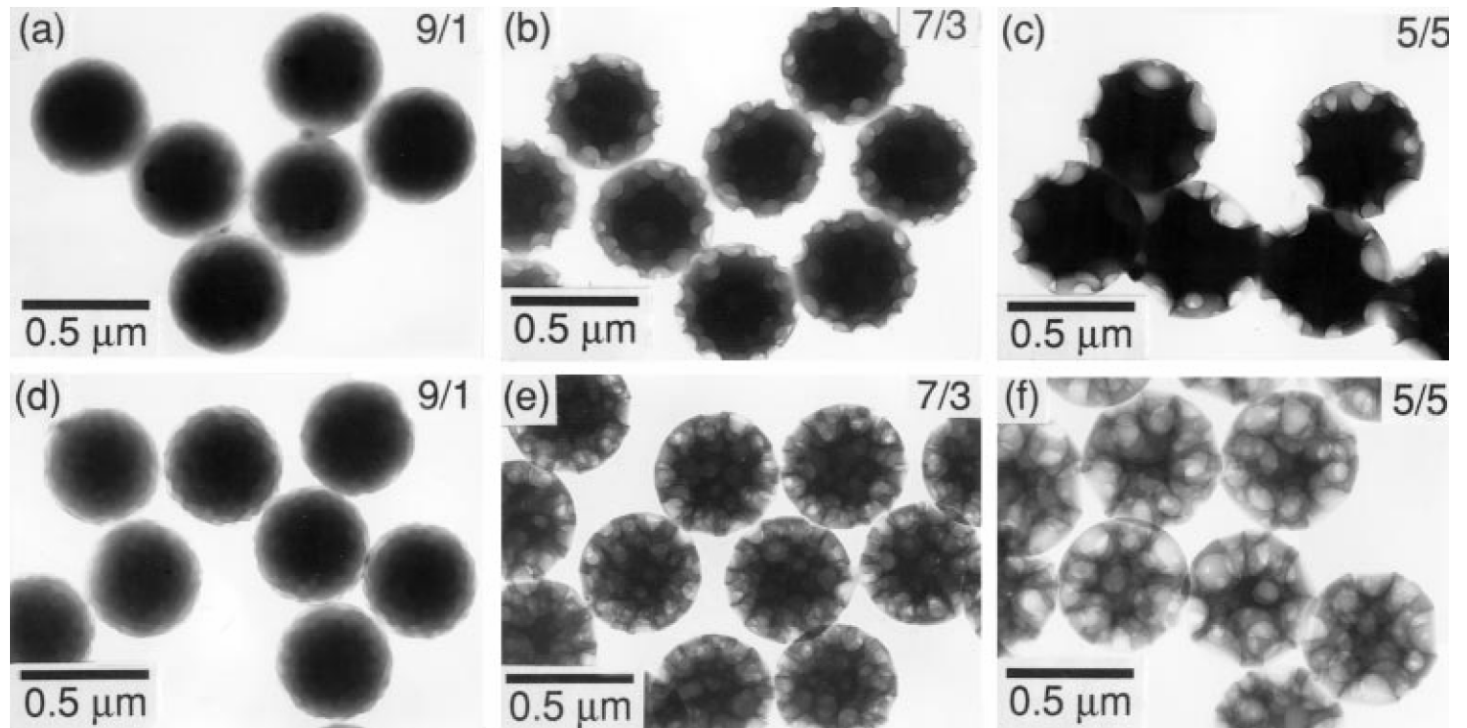
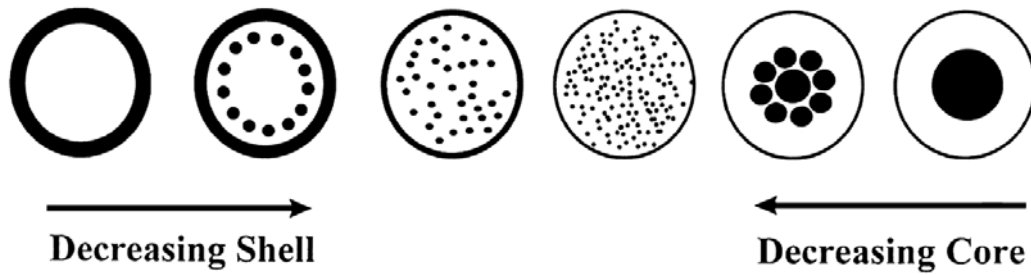


PS  
PBMA



PS/PBMA composite particles

# Penetration depth



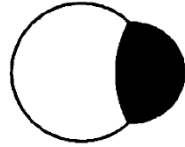
PS/PBA composite particles



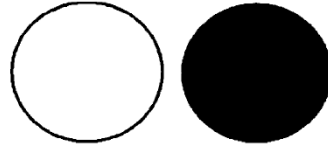
# Entropy and Spreading coefficient effects



Complete engulfment  
 $S_1 < 0, S_2 < 0, S_3 > 0$

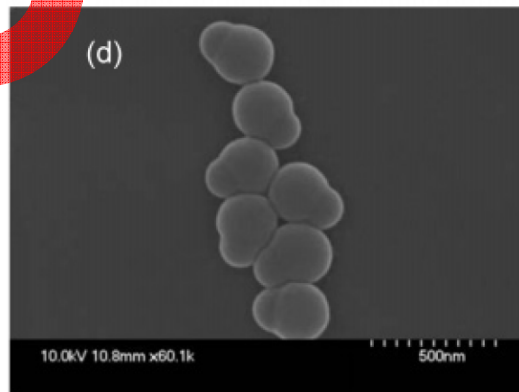
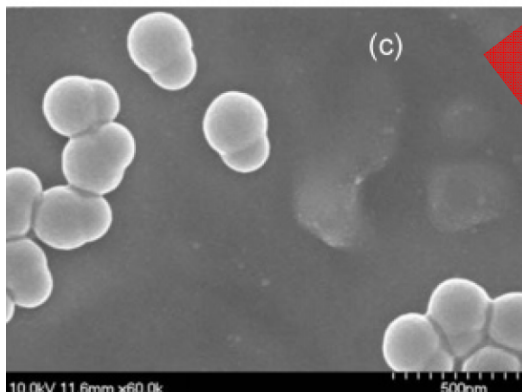
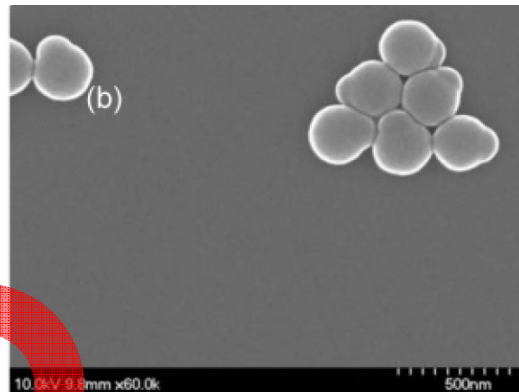
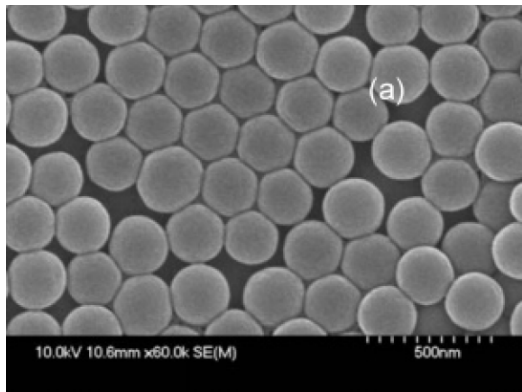


Partial engulfment  
 $S_1 < 0, S_2 < 0, S_3 < 0$



No engulfment  
 $S_1 < 0, S_2 > 0, S_3 < 0$

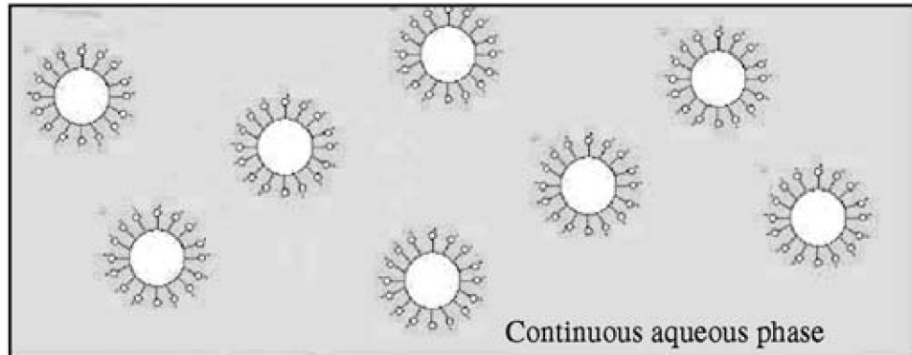
$$S_i = \gamma_{jk} - (\gamma_{ij} + \gamma_{ik})$$



Hydrophilicity of seed particles increases



# Miniemulsion polymerization



## Features

- Heterogeneous system
- Droplet nucleation (no micelles in water)
- Initiation by radial entry (prepolymeric or oligomeric)
- Fairly monodisperse

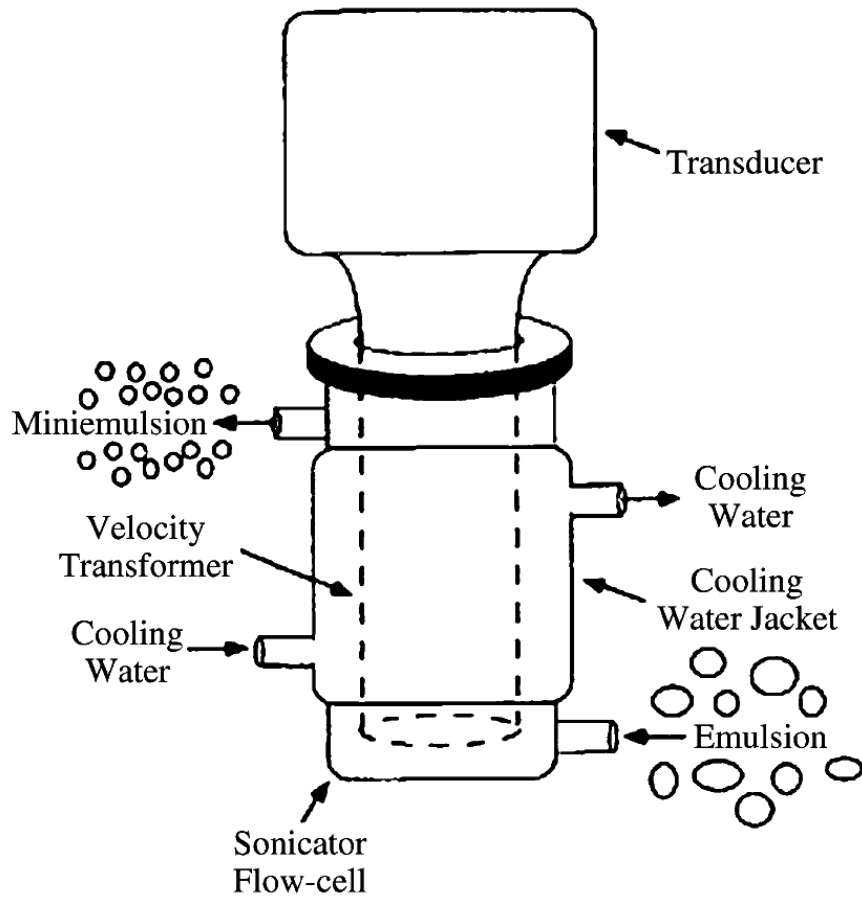
## Role of costabilizers

- Highly insoluble in water; hexadecane, cetyl alcohol, polymers...
- Retarding monomer diffusion from the smaller droplets to the larger
- Preventing Ostwald ripening

## Components

- Monomers
- Water
- Water-soluble initiator
- Surfactant and **"costabilizer"**

# Emulsification

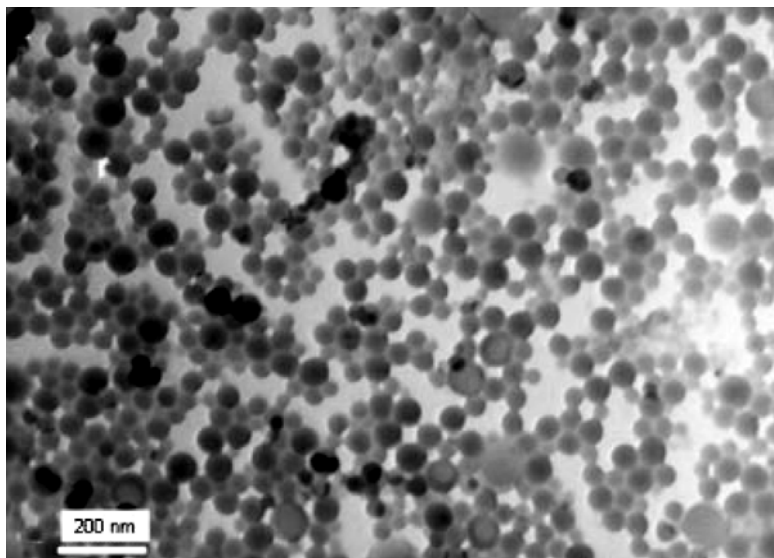


**Mechanically emulsified**

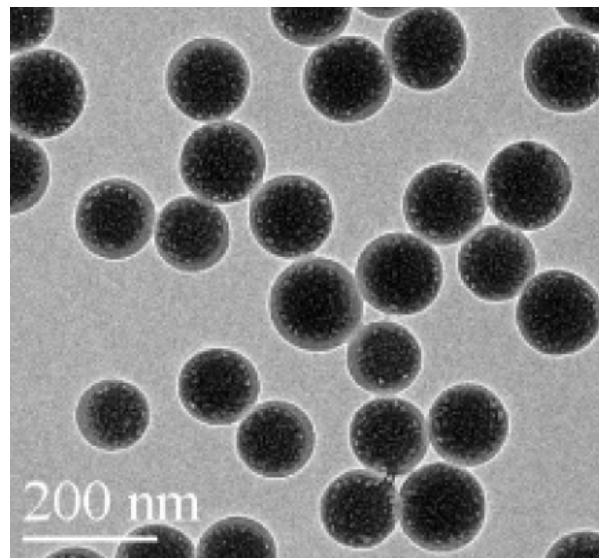


**Macroemulsion and miniemulsion  
after 3 hours**

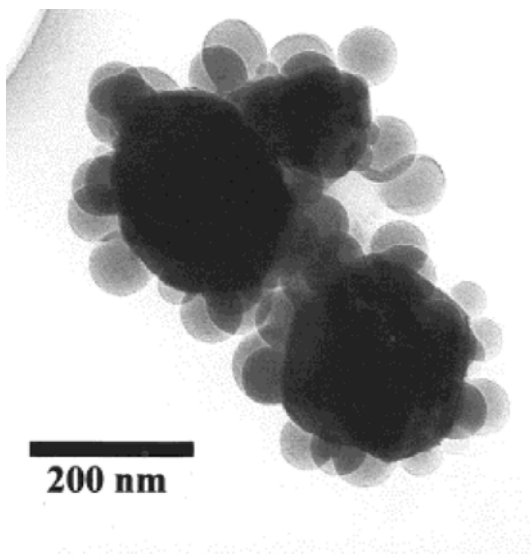
# Polymer particles



PS particles

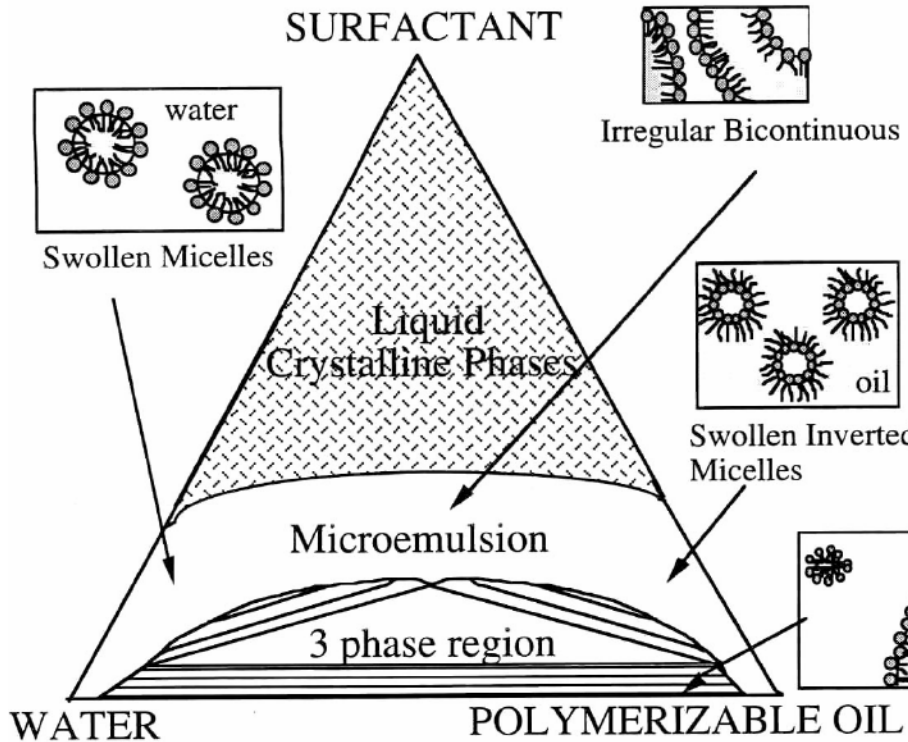


SiO<sub>2</sub>@PS particles



TiO<sub>2</sub>@PS particles

# Microemulsion polymerization



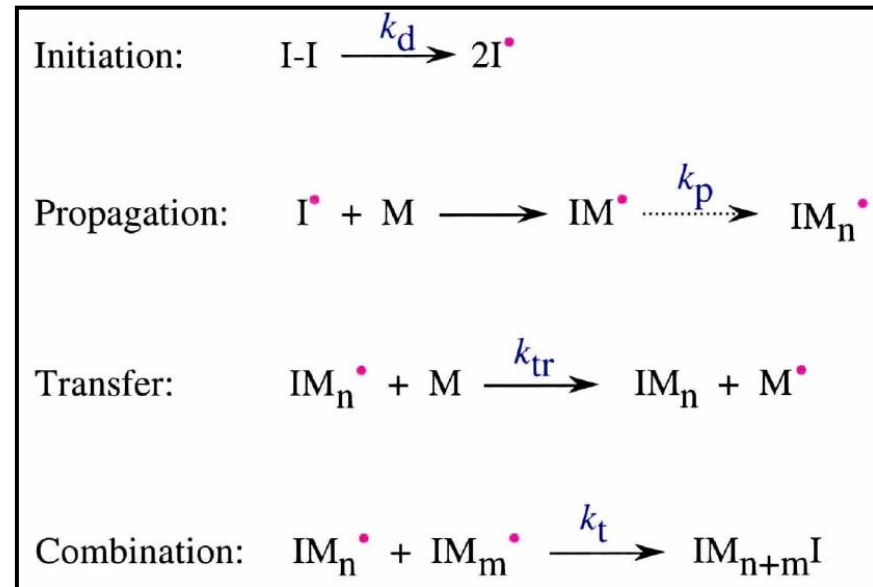
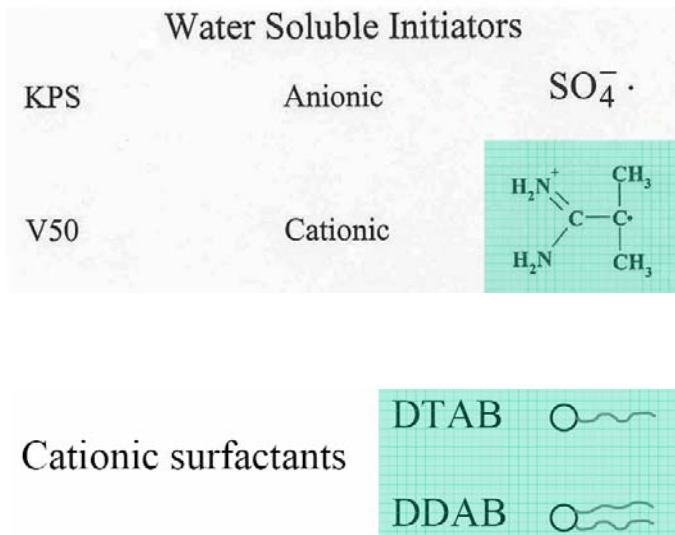
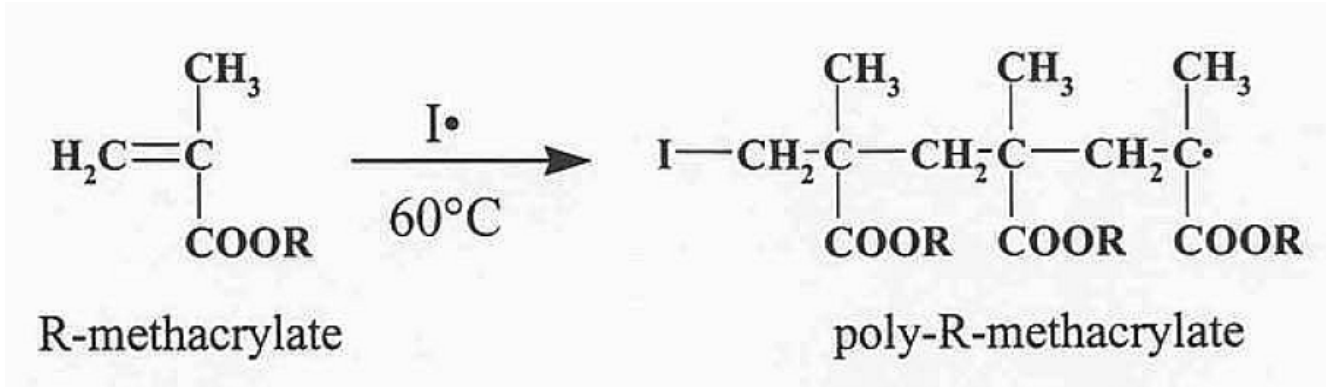
## Microemulsion

- Homogeneous; micelle dispersion
- Spontaneous
- Transparent
- Swollen micelles, less than 20 nm in diameter

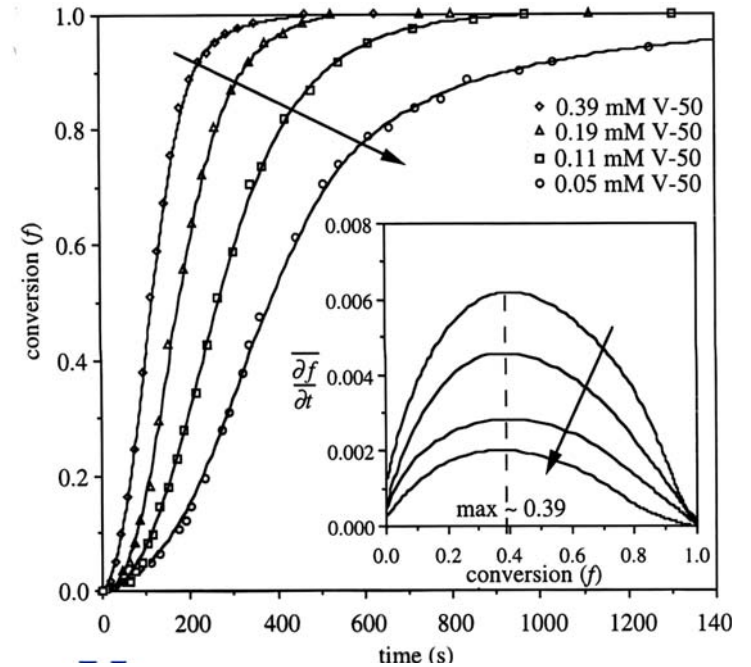
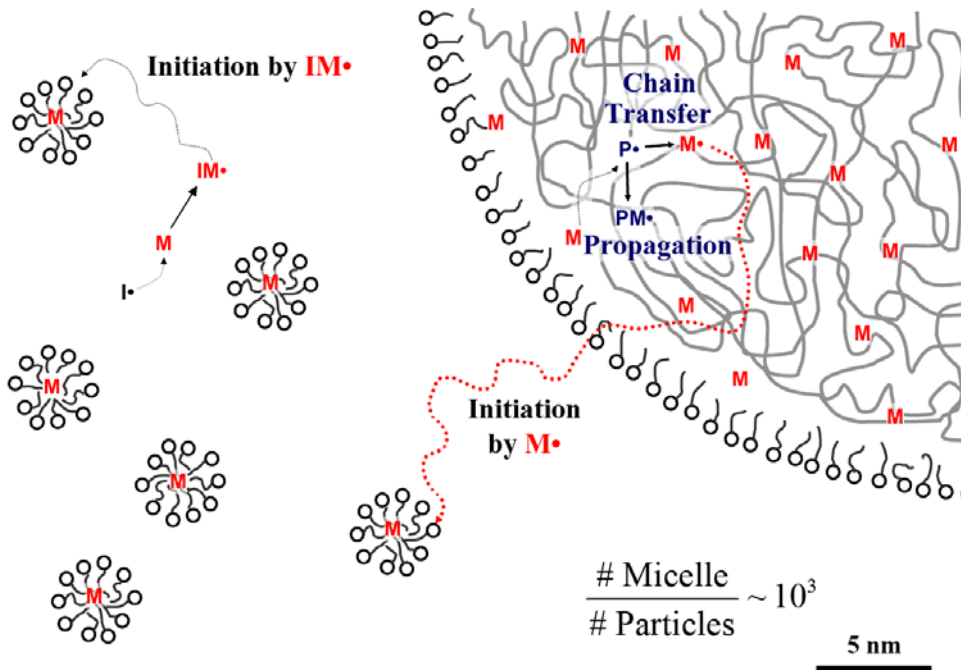
## Microemulsion polymerization

- Tacticity
- Knotting
- A high MW, more than 20 million Dalton

# Polymerization

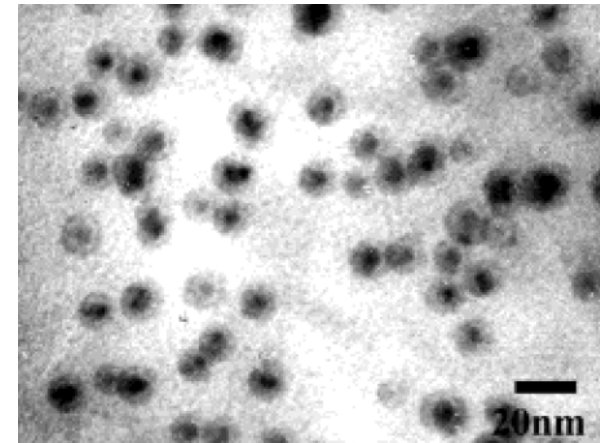


# Polymerization process



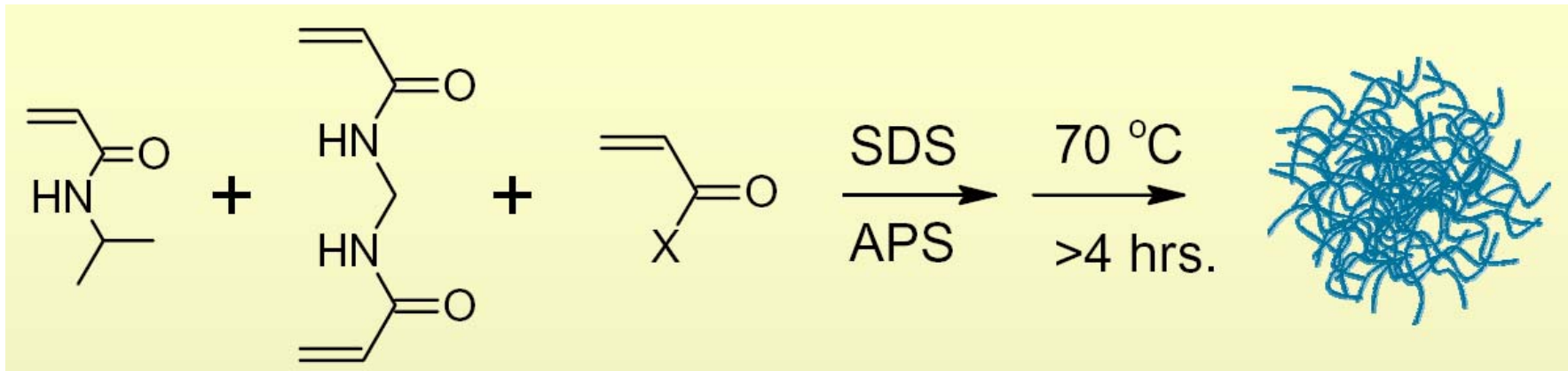
- Rapid polymerization
- 100% conversion
- Rate profile: parabolic
- Maximum rate  $\sim 40\%$

PMMA/PS  
nanoparticles

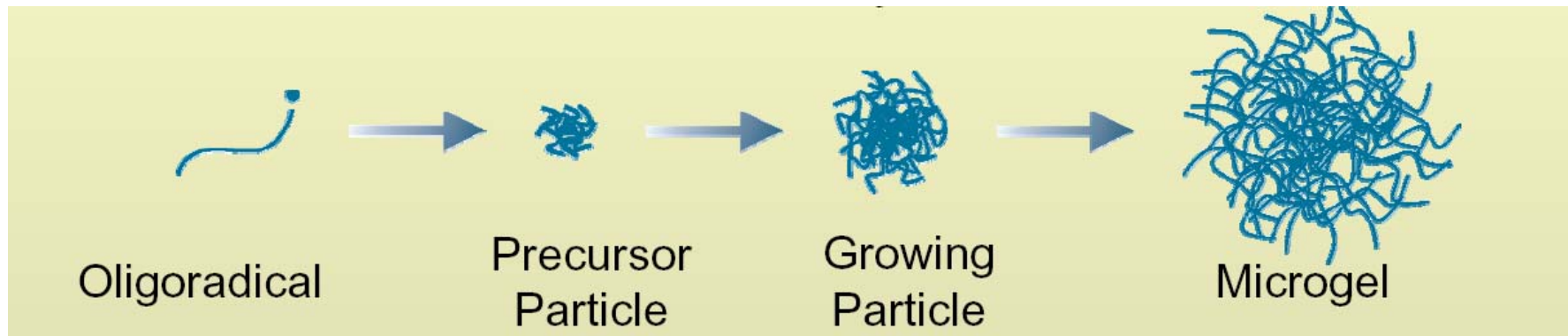




# Precipitation polymerization

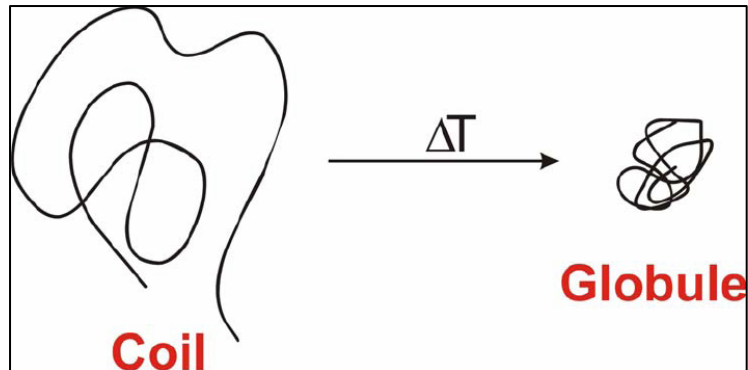


A basic composition



A nucleation and growth process

# Properties of this technique



## Fundamentals

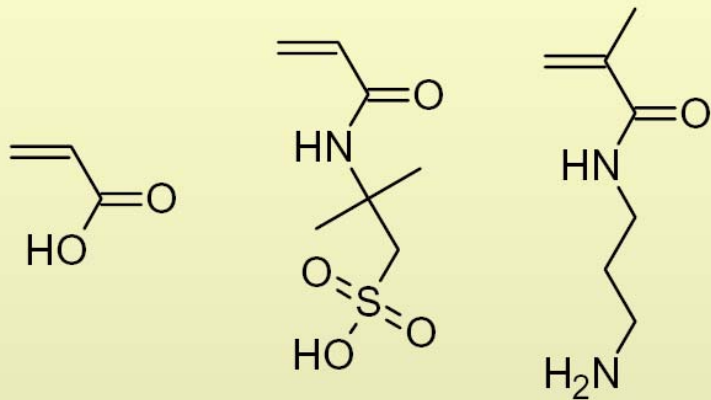
- Homogeneous system
- Nucleation site is water
- A nucleation and growth
- Polymerization time: short (several hours)
- From hundreds of nanometers to 1 or 2 micrometers
- Very monodisperse
- Surfactant-free
- Crosslinking is possible

## Components

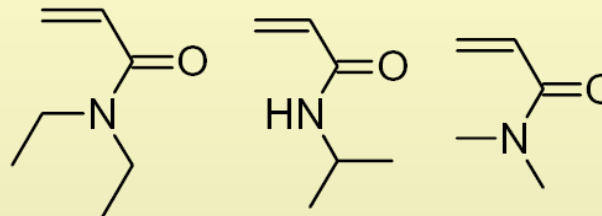
- Monomers (water-soluble)
- Water
- Water-soluble initiator
- Surfactant (if need)



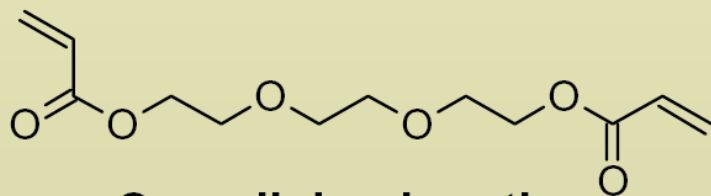
# Designing microgel particles



pH sensitive/polyelectrolytes



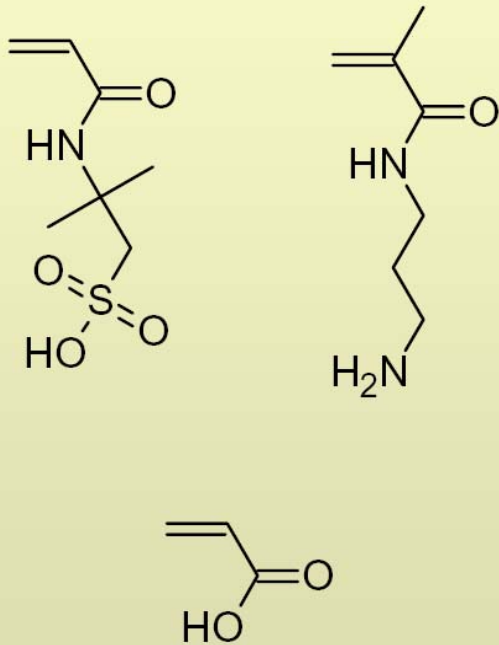
thermosensitive



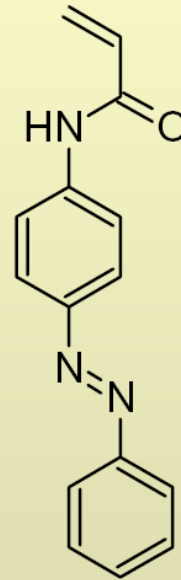
Crosslinker length

# Responsivity and sensitivity

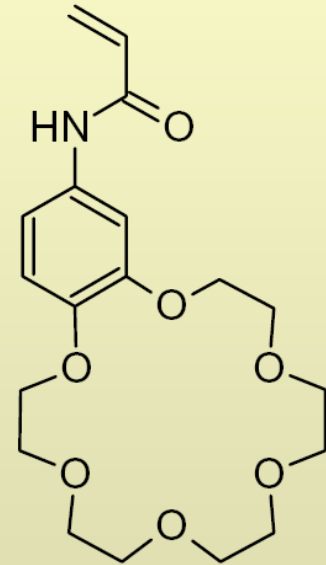
## Electric Field



## Light

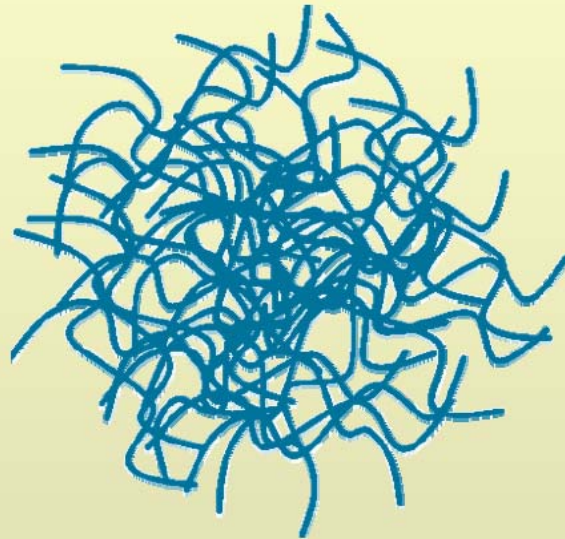


## Metal Ions



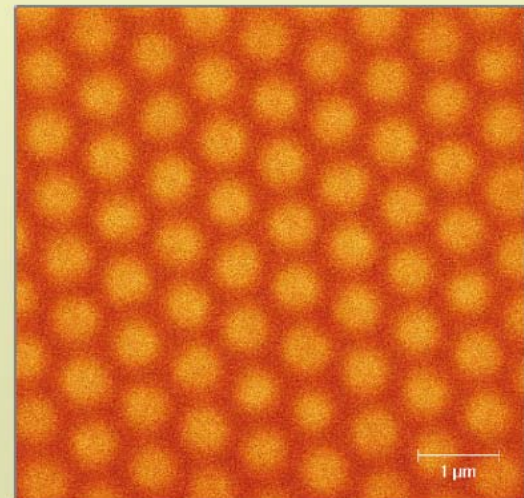
# Microgel characterization

Infinite Spherical  
Network

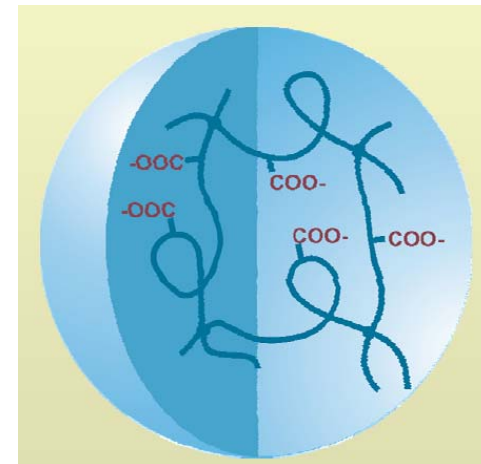
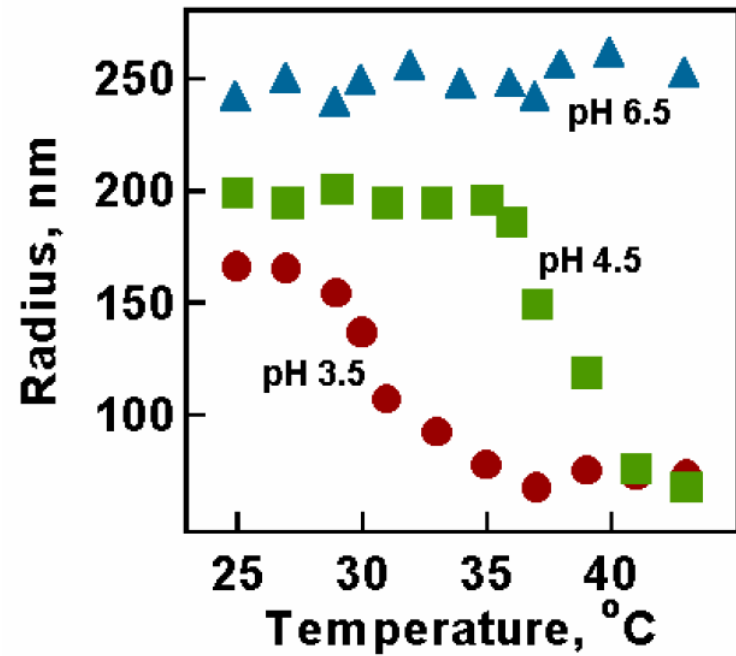
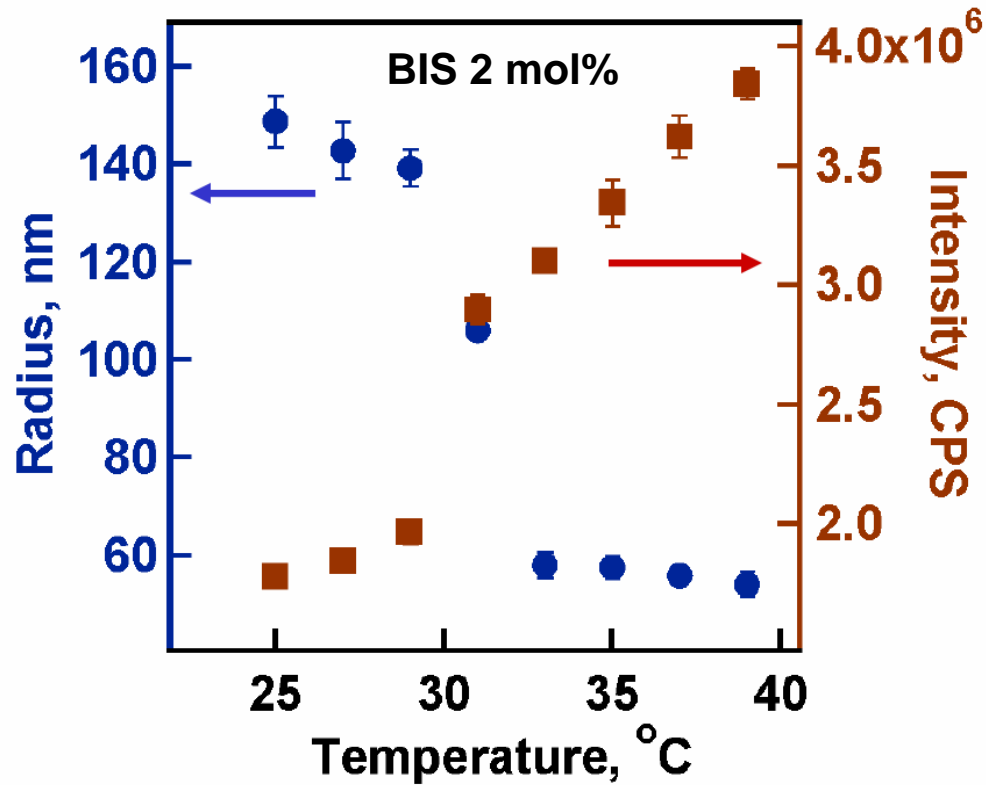


High water content (90-99% v/v)  
a microgel is effectively *all surface area*

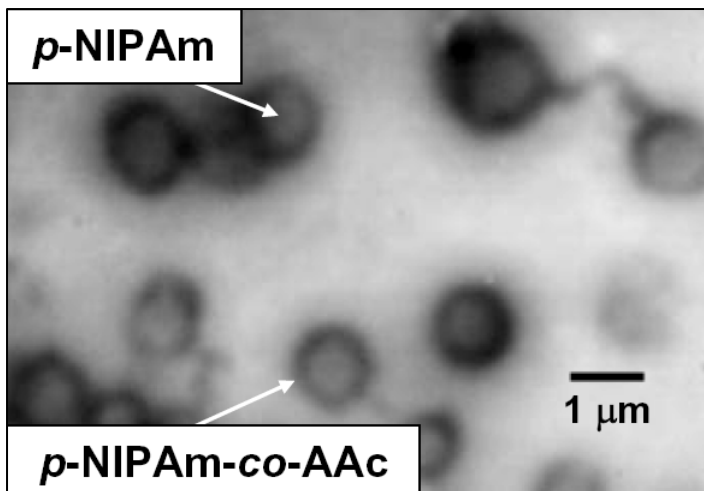
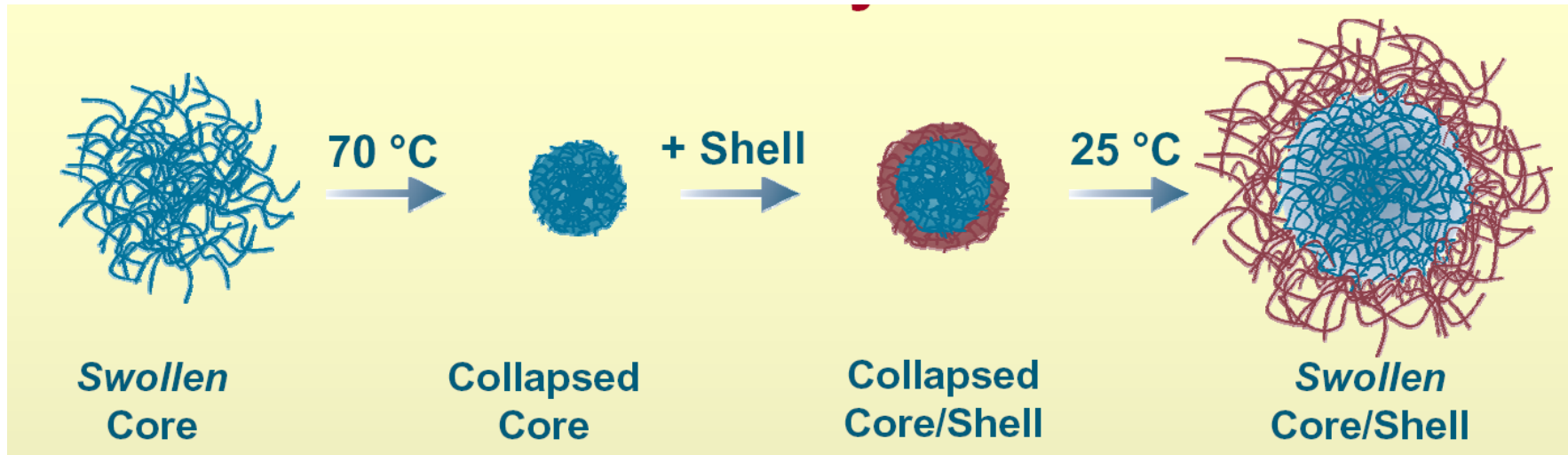
Highly Monodisperse



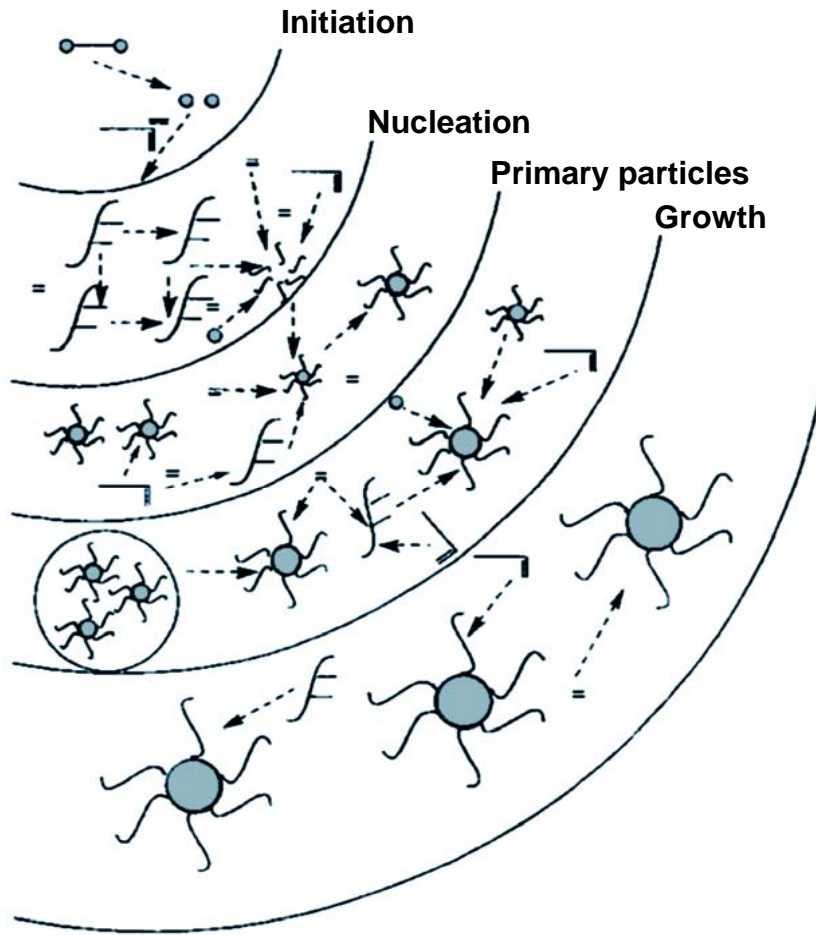
# Phase transitions



# Core-shell microgel particles



# Dispersion polymerization



A dispersion polymerization process

## Fundamentals

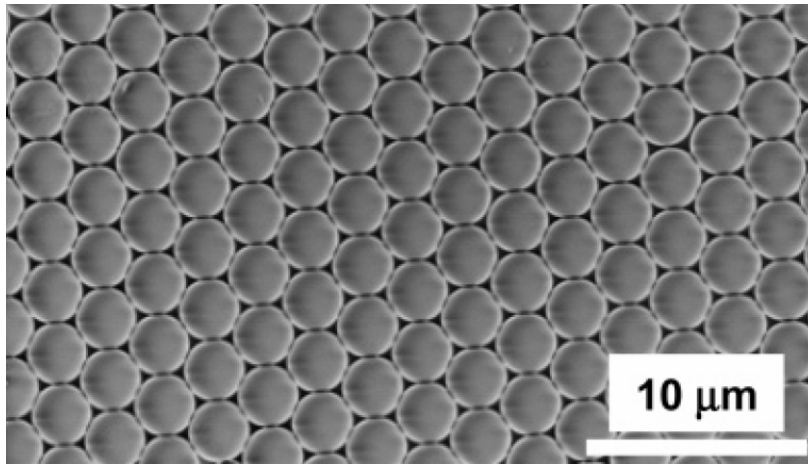
- Homogeneous system
- Nucleation site is media
- A nucleation and growth
- Polymerization time: long (more than 24 hours)
- From a micromerter to 15 micrometers
- Very monodisperse
- Stabilizer is needed
- Crosslinking is impossible

## Components

- Monomers (medium-soluble)
- Alcohol
- Water-insoluble initiator
- Stabilizer; polymer (PVP, PAAc....)

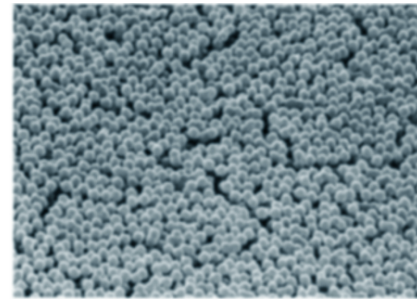


# Monodisperse polymer microparticles

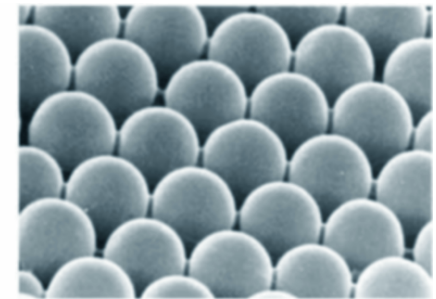


**Monodisperse PS particles**

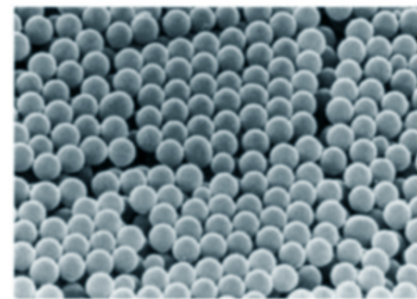
- PVP
- Ethanol medium



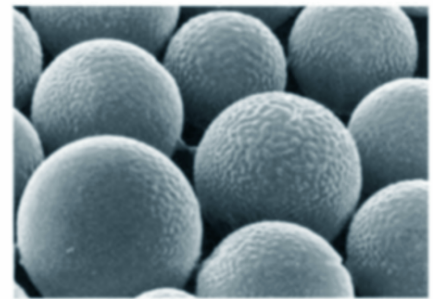
(A)



(C)



(B)

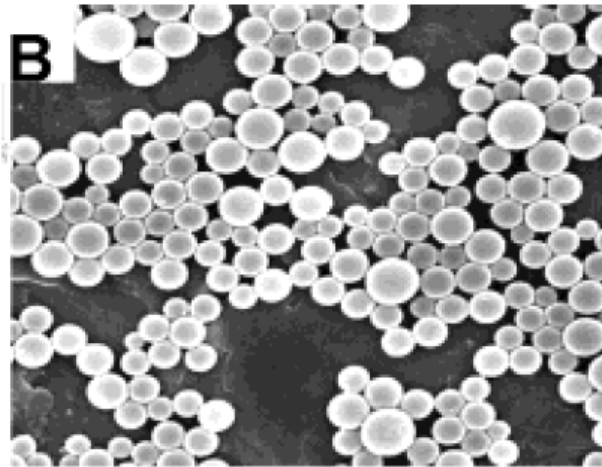
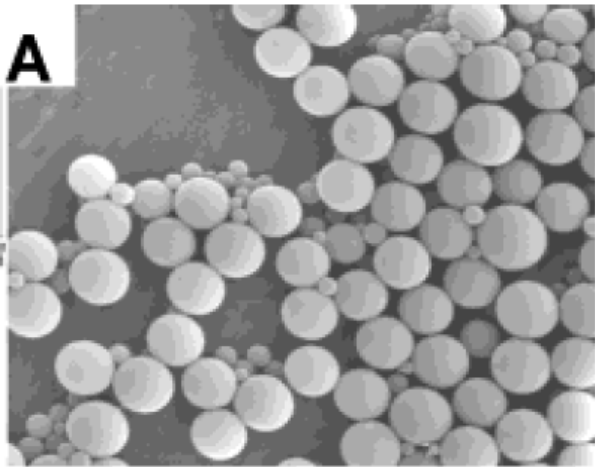


(D) ——— 5μm

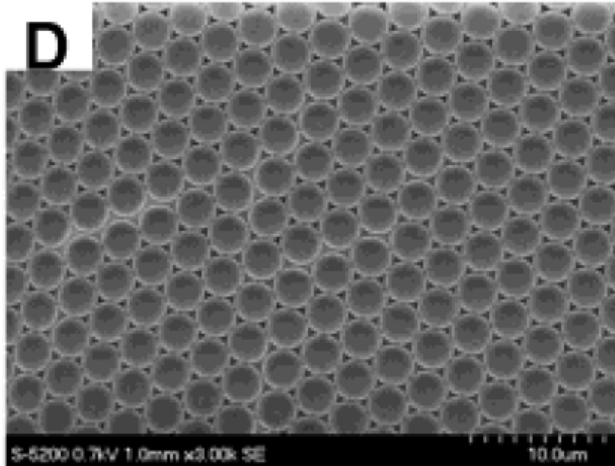
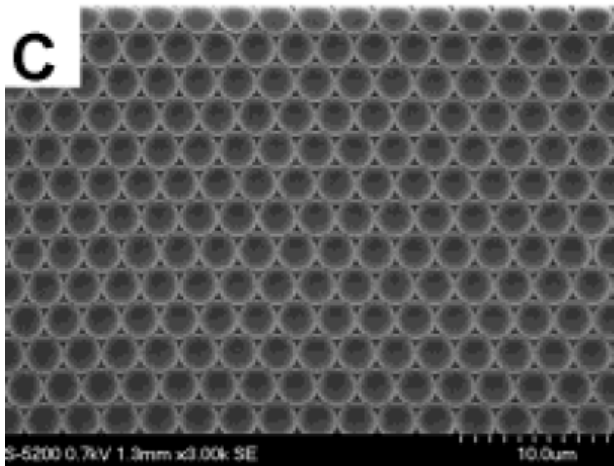
**Poly (HEMA) microparticles**

- Poly (St-b-BD) stabilizer
- 2-butanol/toluene media

# Crosslinking in dispersion polymerization



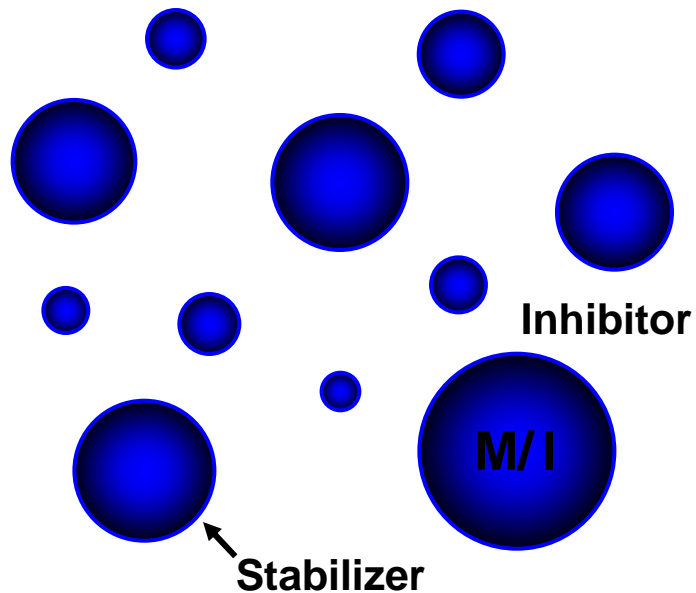
Conventional method;  
(a) 1% dye; (b) 0.35% DVB



Two-staged method;  
(c) 1% DVB; (d) 1% EGDMA

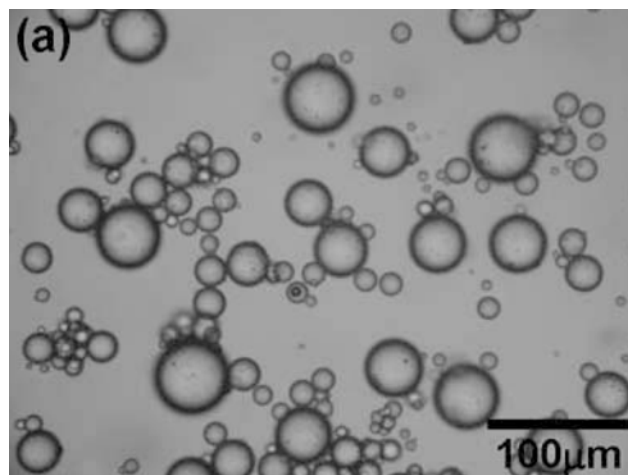


# Suspension polymerization

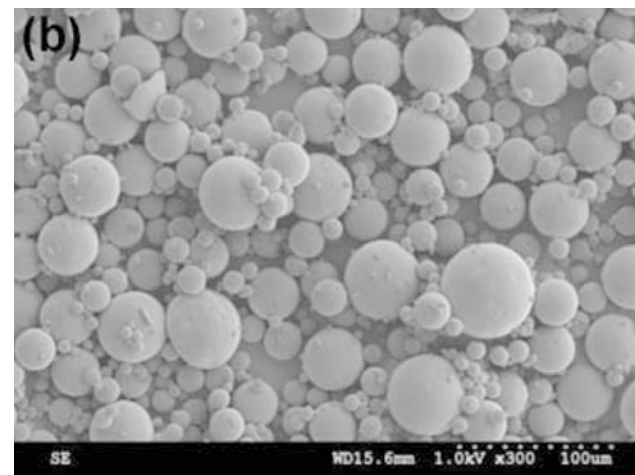


## Fundamentals

- A heterogeneous system
- Nucleation site is monomer phase (droplet)
- Polymerization time: long (more than 10 h)
- From 1 micrometer to thousands of micrometers
- Very polydisperse



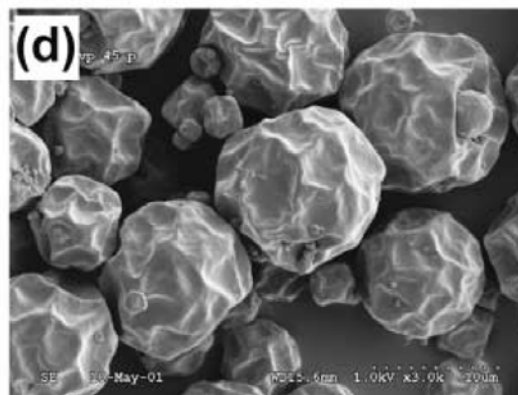
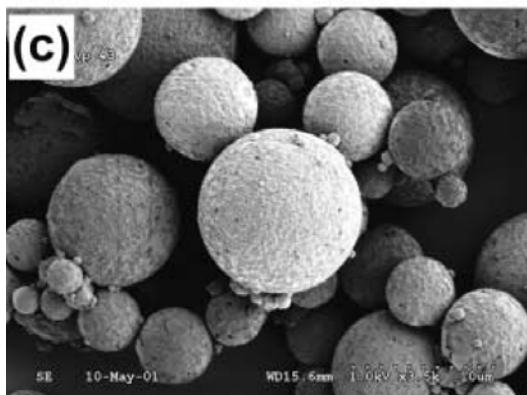
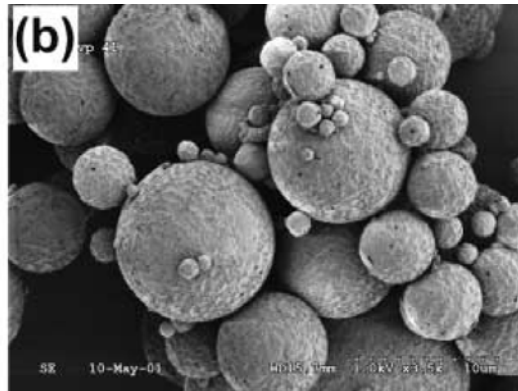
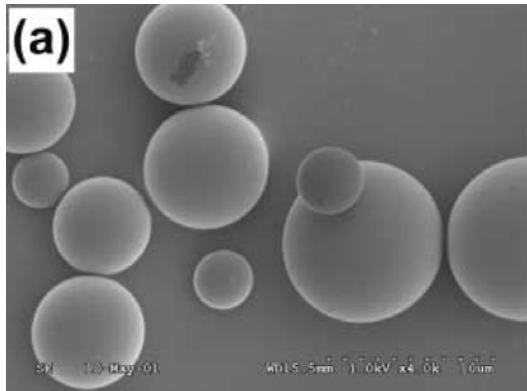
MMA droplets



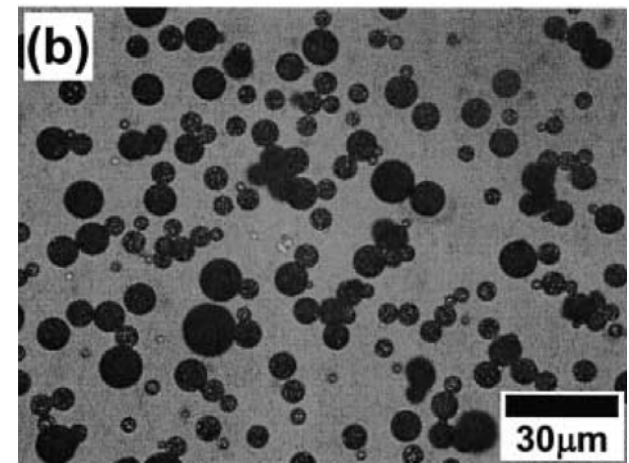
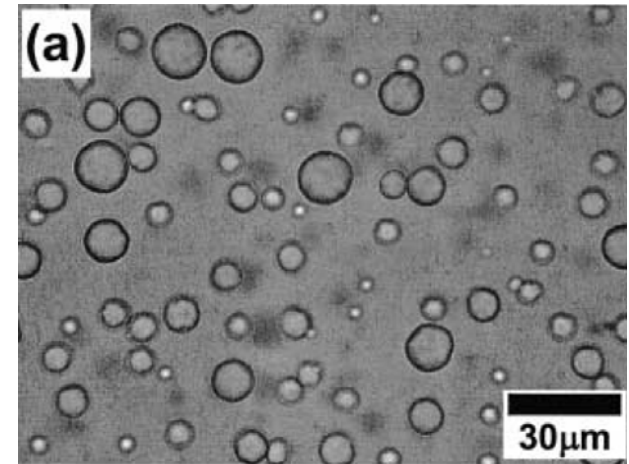
PMMA particles



# Inorganic/polymer composite particles

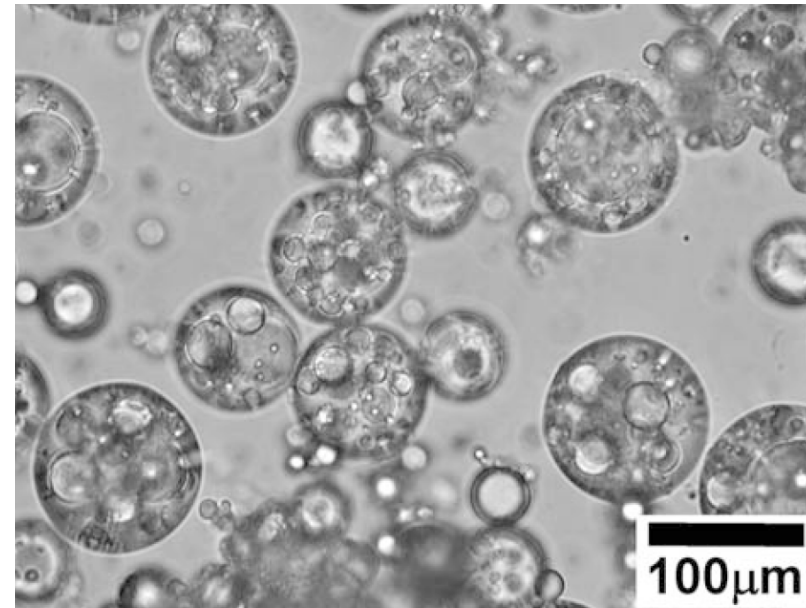
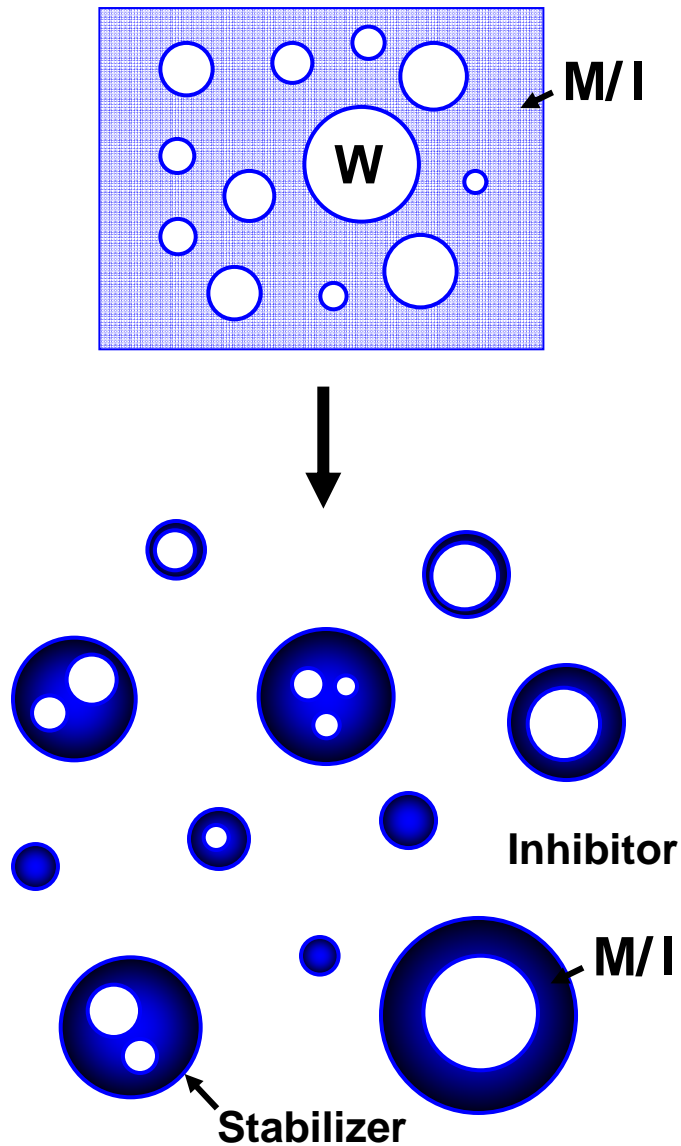


**TiO<sub>2</sub>/PMMA composite particles  
(30 nm TiO<sub>2</sub>)**



**Light blocking**

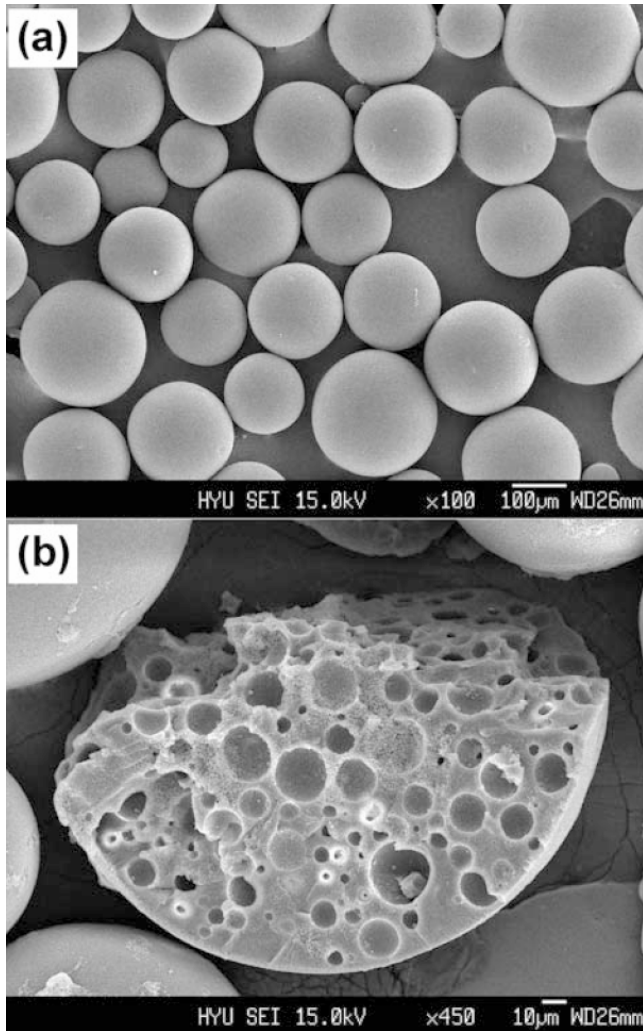
# W/O/W double emulsion polymerization



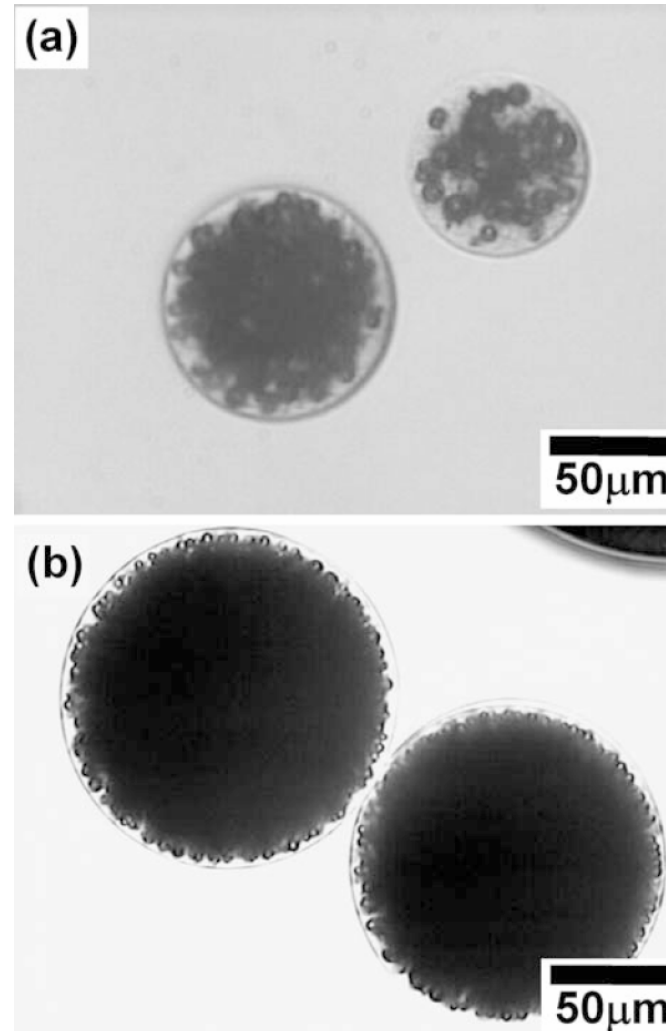
W/O/W double emulsions  
(Oil is the mixture of MMA and AIBN)



# Hollow polymer particles

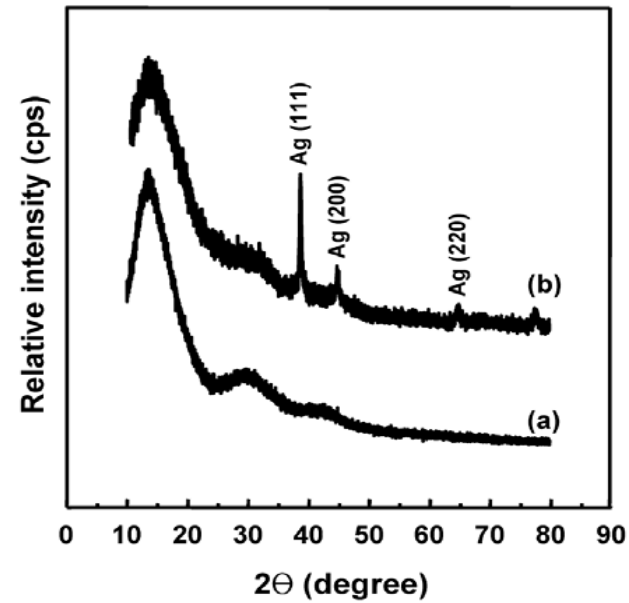
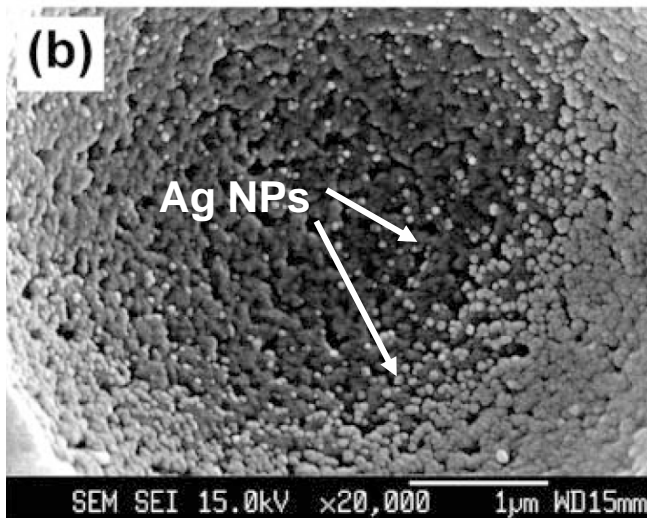
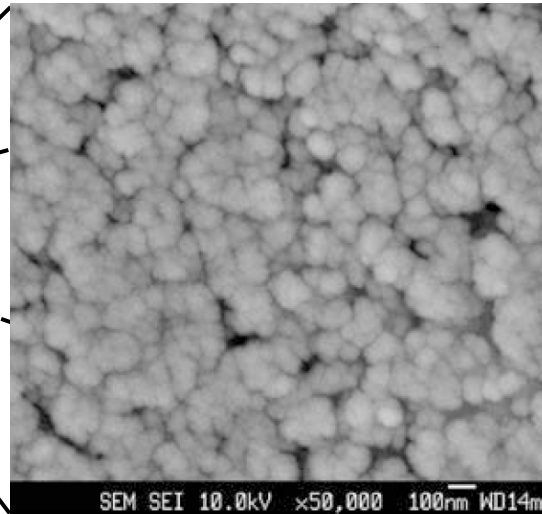
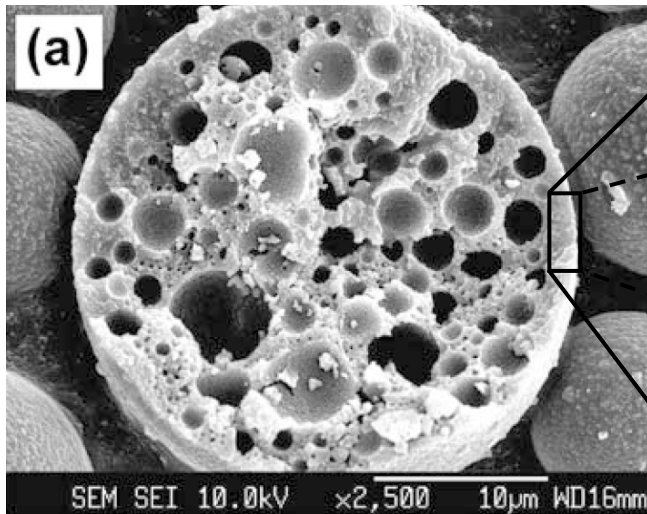


PMMA hollow particles

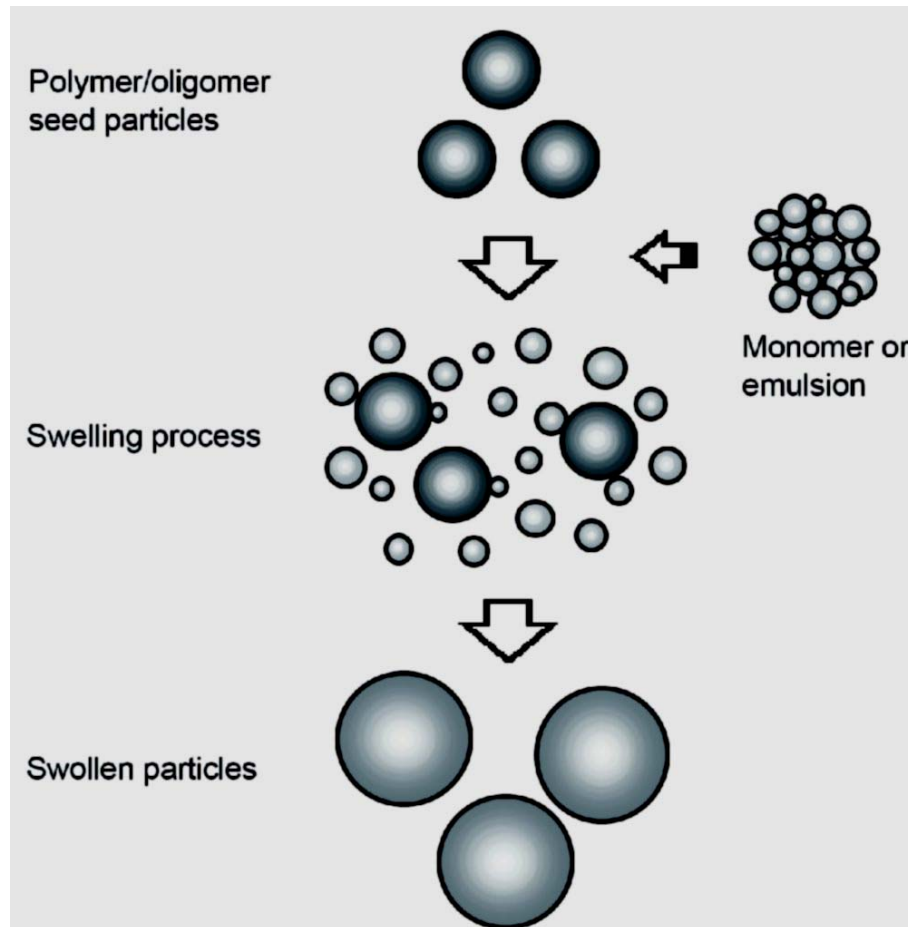


Dark pores in toluene

# Ag NP in pores



# Seeded swelling & polymerization



## Fundamentals

- A heterogeneous system
- Nucleation site is monomer phase (droplet)
- A swelling and polymerization
- Polymerization time: long (more than 10 h)
- From tens of nanometers to thousands of micrometers
- Very monodisperse
- Stabilizer: PVA, PAAc...
- Crosslinking is possible

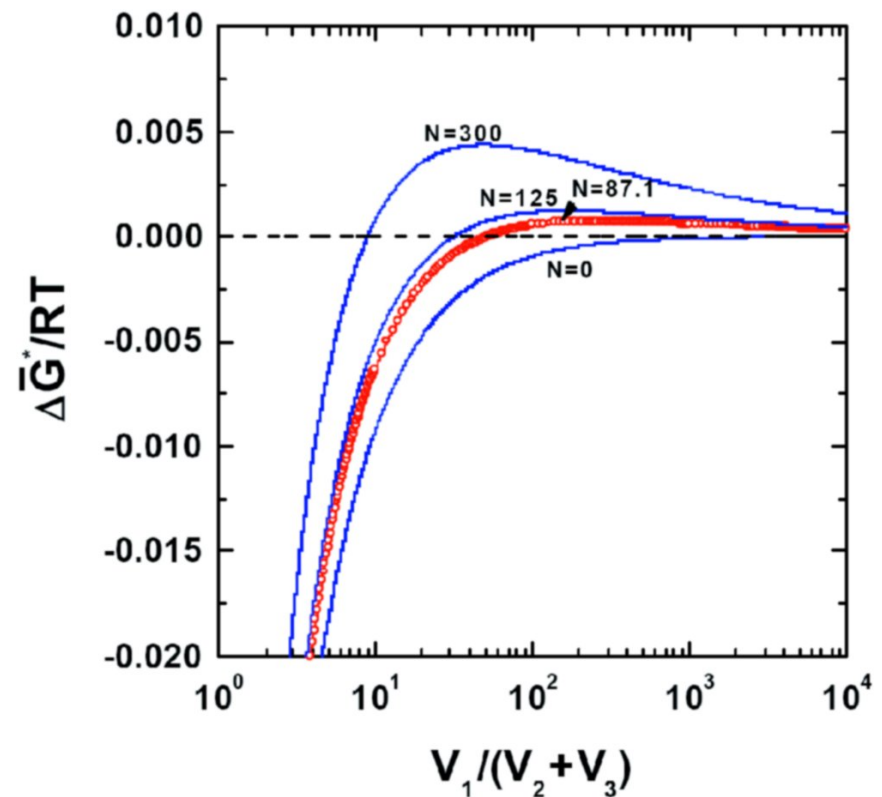
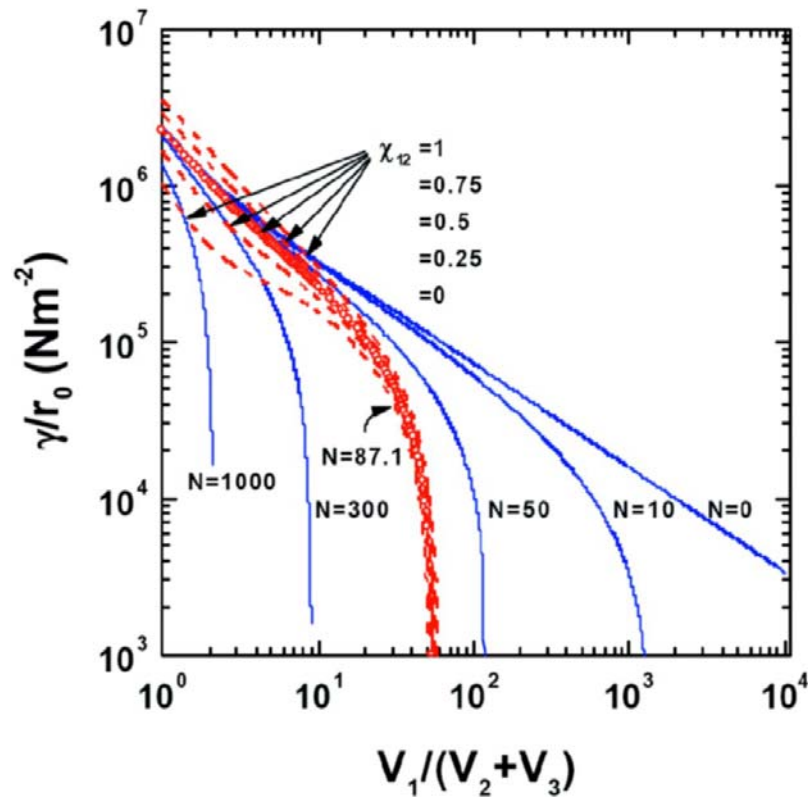
## Components

- Seed particles
- Water
- Monomers (with initiator)
- Stabilizer

# Swelling process

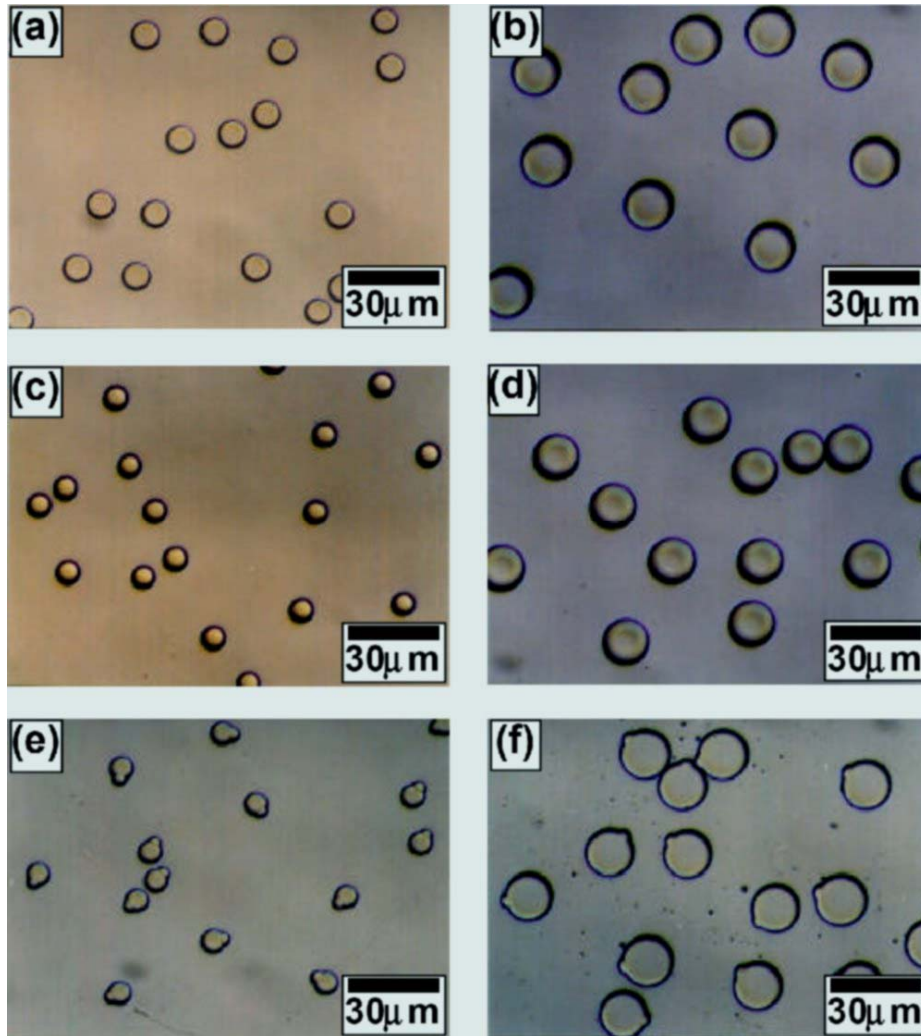
$$\Delta\bar{G} = RT \left[ \ln \phi_1 + \left(1 - \frac{1}{j}\right) \phi_2 + \phi_2^2 \chi + \frac{2\bar{V}_1 \gamma}{rRT} \right]$$

$$\begin{aligned} \Delta\bar{G}^* = RT & \left[ \ln \phi_1 + \left(1 - \frac{1}{j_2}\right) \phi_2 + \phi_3 + \phi_2^2 \chi_{12} + \phi_3^2 \chi_{13} \right. \\ & + \phi_2 \phi_3 \left( \chi_{12} + \chi_{13} - \frac{\chi_{23}}{j_2} \right) + \frac{2\bar{V}_1 \gamma}{rRT} \\ & \left. + N\bar{V}_1 \left( \phi_3^{1/3} - \frac{\phi_3}{2} \right) \right] \end{aligned}$$

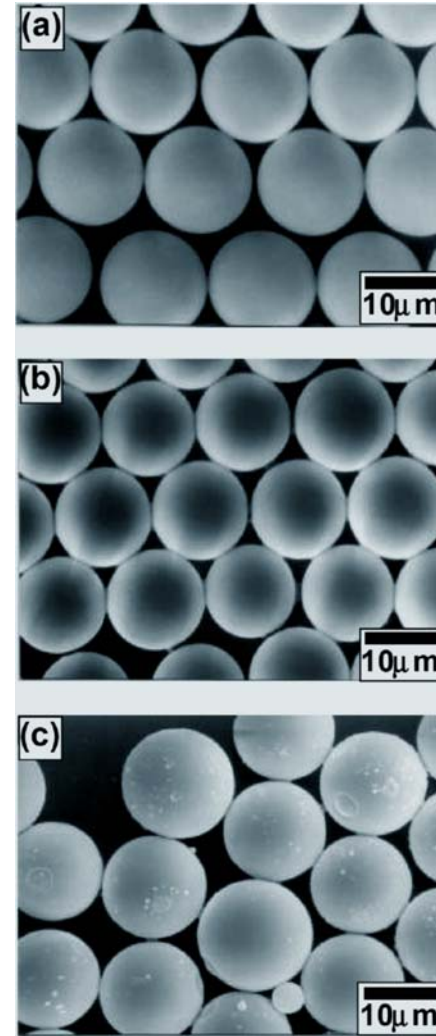




# Swelling and polymerization

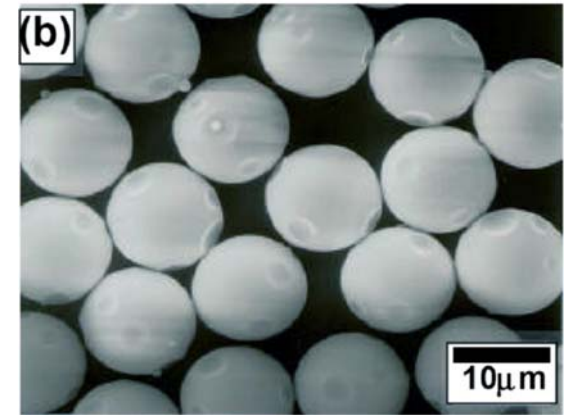
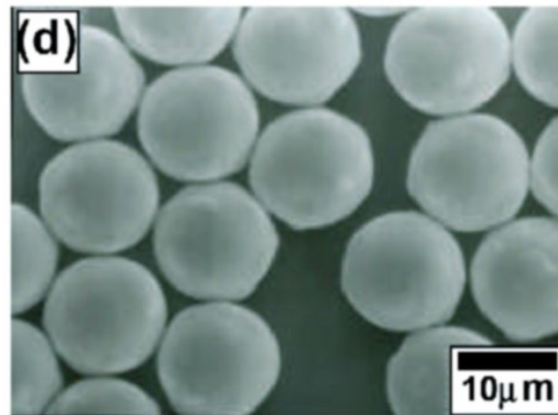
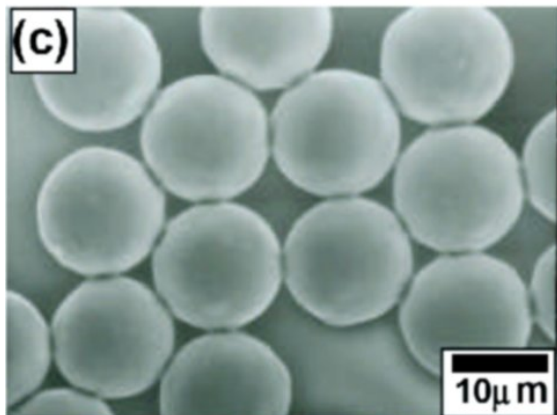
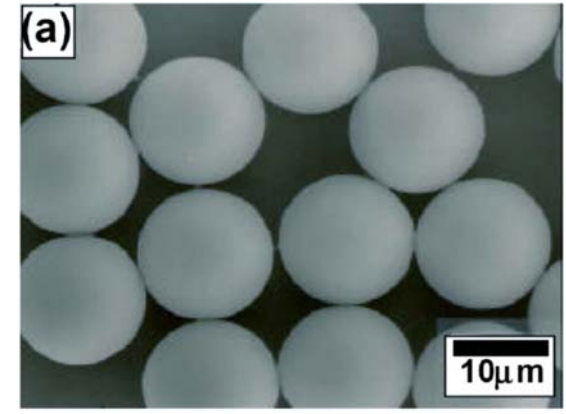
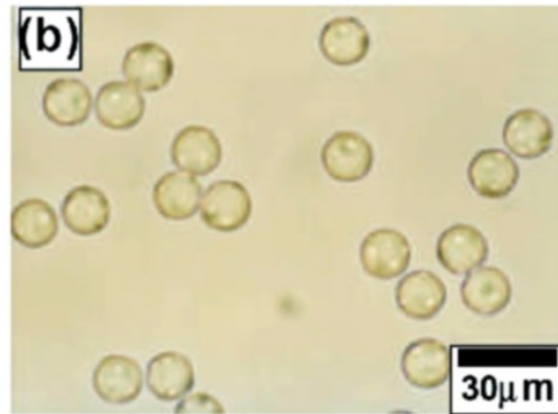
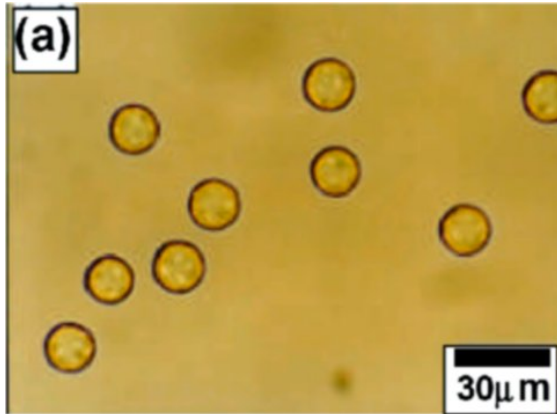


**Monomer swellings**  
(after 30 sec and 2h)



**PS particles**

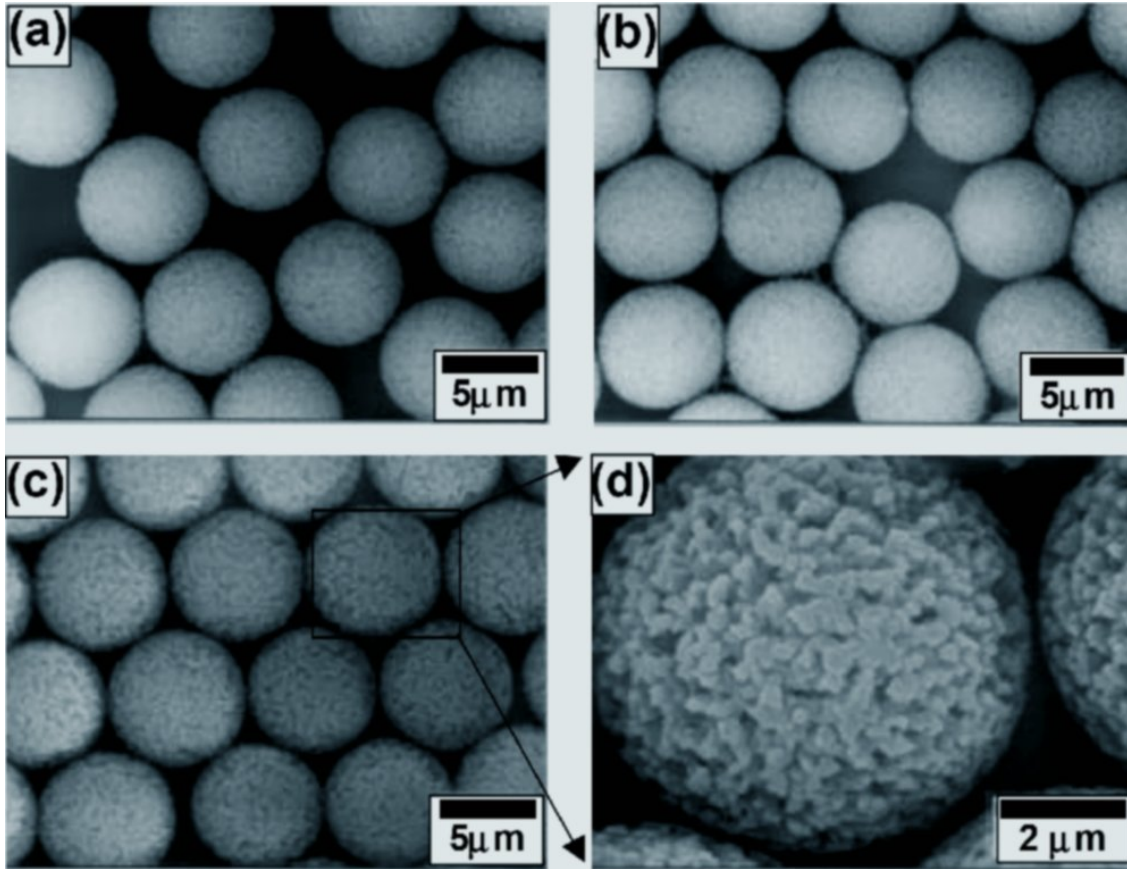
# Composite particles



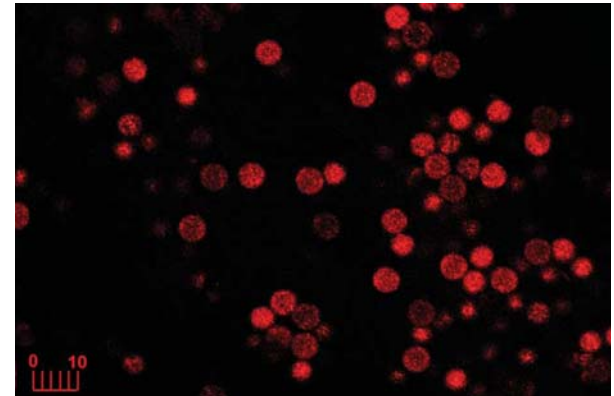
PS/PGMA composite particles

PS/PCMS composite particles

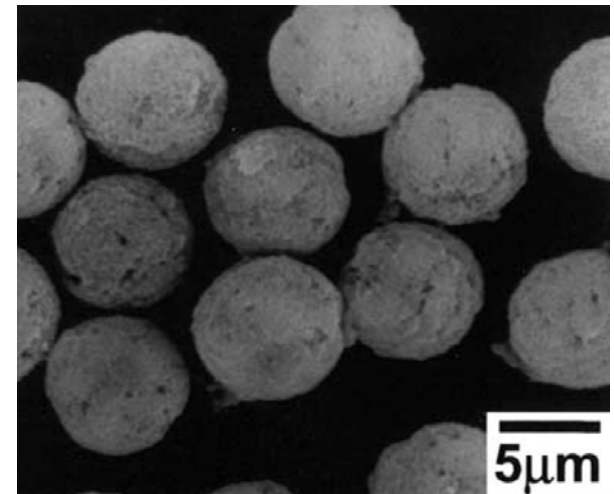
# Porous particles



Porous PS particles



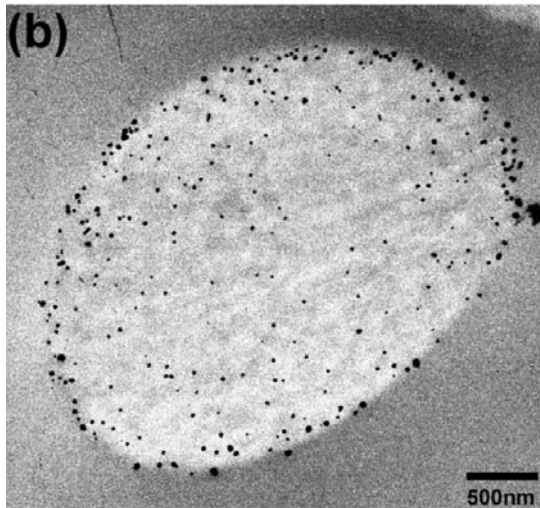
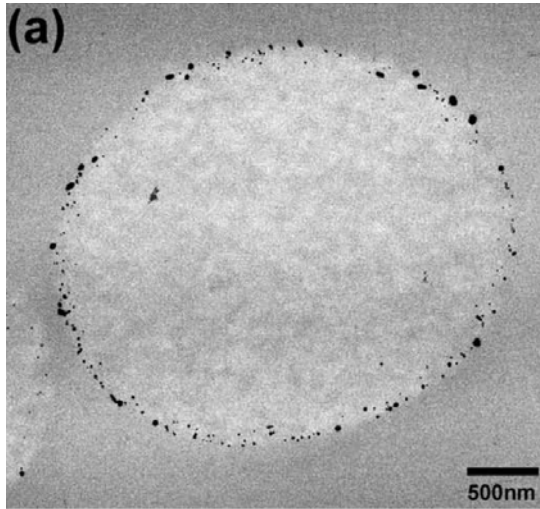
Resveratrol loaded particles



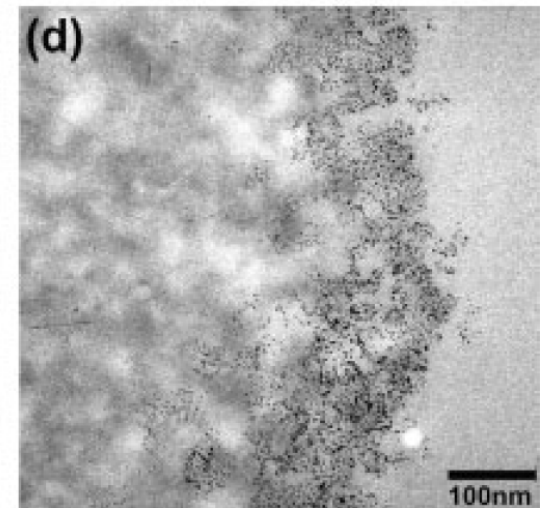
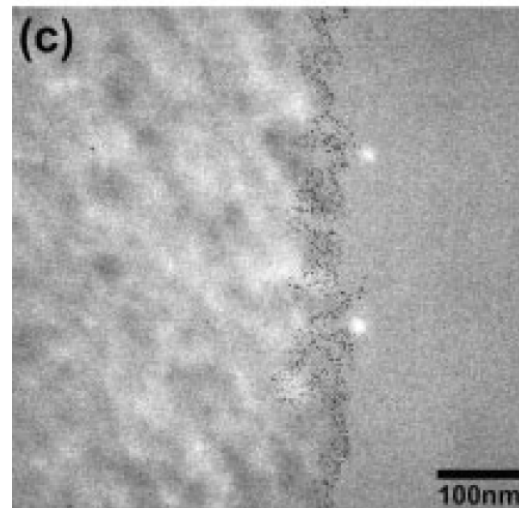
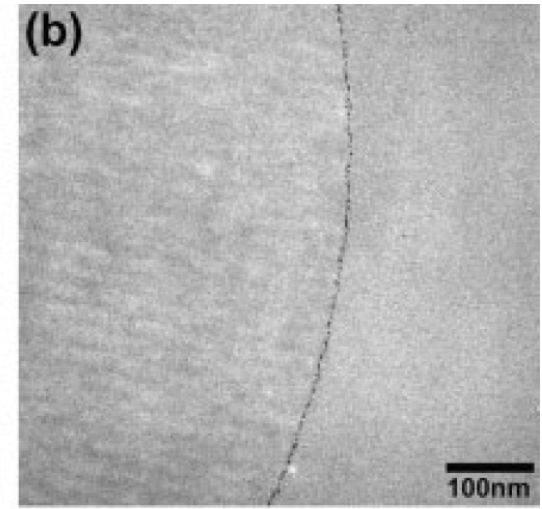
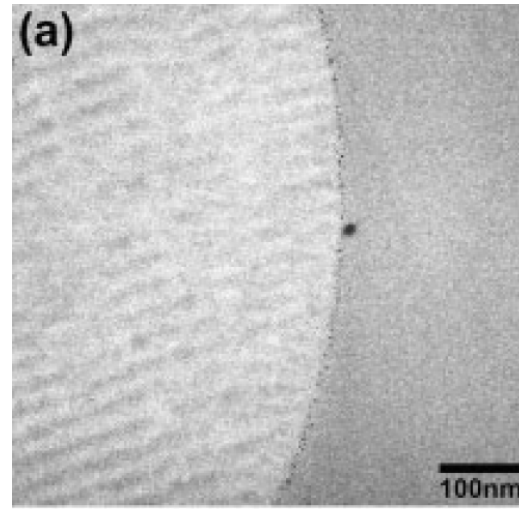
Conducting particles  
(PS/polyaniline)



# Metal/polymer composite particles

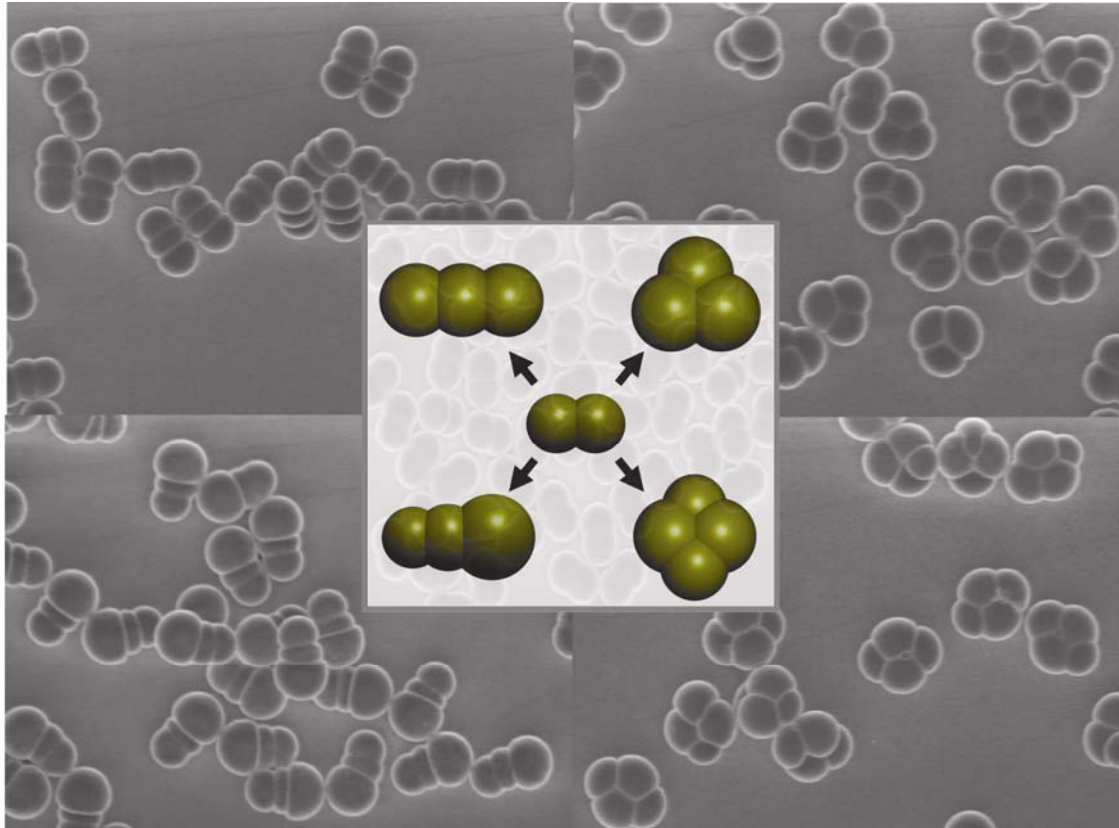


Ag/poly(EGDMA) particles

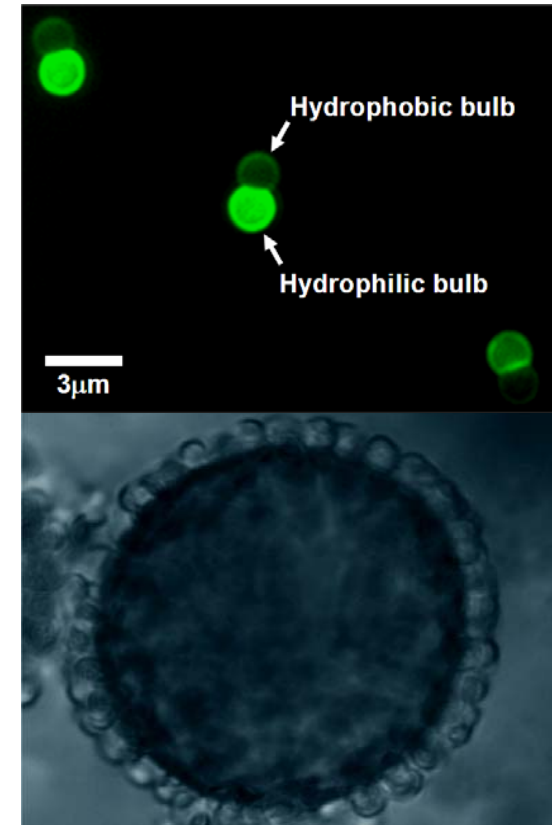


Au/poly(EGDMA) particles

# Non-spherical particles



Controlling shape anisotropy



Controlling chemical anisotropy

“Combination of dispersion polymerization and seeded polymerization”